

Report Sheet: Compounds and Their Formulas

Date _____

Section _____

Instructor _____

Name _____

Team _____

Pre-Lab Study Questions

1. Where are the valence electrons in an atom?

Period	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
Group	1	2	13	14	15	16	17	18	3	4	5	6	7	8	9	10	11	12

2. How are positive and negative ions formed?

Period	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
Group	1	2	13	14	15	16	17	18	3	4	5	6	7	8	9	10	11	12

Ionic Compounds and Formulas

3. How do subscripts represent the charge balance of ions?

Period	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
Group	1	2	13	14	15	16	17	18	3	4	5	6	7	8	9	10	11	12

4. Why are electrons shared in covalent compounds?

Period	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
Group	1	2	13	14	15	16	17	18	3	4	5	6	7	8	9	10	11	12

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**New York State Agricultural
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General, Organic, and Biological Chemistry Study Guide and Selected Solutions Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems *Laboratory Manual for Principles of General Chemistry* J. A. Beran, Mark Lassiter, 2022-08-16 Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write textbooks textbooks don't write experiments A student's scientific literacy grows when experiences and observations associated with the scientific method are encountered Further experimentation provides additional cause effect observations leading to an even better understanding of the experiment The 11th edition's experiments are informative and challenging while offering a solid foundation for technique safety and experimental procedure The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit An Additional Notes column is included in each experiment's Report Sheet to provide a space for recording observations and data during the experiment Continued emphasis on handling data is supported by the Data Analysis section **Laboratory Manual for Principles of General Chemistry** Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures *A Laboratory Book of Elementary Organic Chemistry* Alexander Lowy, Wilmer E. Baldwin, 1926 **Modern Experimental Chemistry** George W. Jr. Latimer, 2012-12-02 Modern Experimental Chemistry provides techniques of qualitative analysis that reinforce experiments on ionic equilibria This book includes the determination of water in hydrated salts identification of an organic compound after determining its molecular weight and nonaqueous titration of a salt of a weak acid The calculation of chemical stoichiometry calculation of thermodynamic properties by determining the change in equilibrium with temperature and chromium chemistry are also covered This compilation contains enough experiments for classes which have six hours of laboratory two 3 hour meetings per week to last two semesters This publication is intended for chemistry students as an introductory manual to chemistry laboratory Laboratory Experiments for Brown and LeMay, Chemistry, the Central Science John Henry Nelson, 1985

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