#### CHEMISTRY

#### Overall Grade Boundaries

#### Standard Level

| Grade:      |       | 2     | 3     | 4     | 5     | - 6   | 7      |
|-------------|-------|-------|-------|-------|-------|-------|--------|
| Mark range: | 0~1.5 | 16-28 | 29-38 | 39-51 | 52-63 | 64-76 | 77-100 |

## Standard Level Paper 1

#### Component Grade Boundaries

| Grade:      | 1   | 2    | 3     | 4     | 5     | 6     | 7     |
|-------------|-----|------|-------|-------|-------|-------|-------|
| Mark range: | 0-7 | 8-10 | 11-13 | 14-17 | 18-21 | 22-25 | 26-30 |

#### **General Comments**

This paper consisted of 30 questions on the Subject Specific Core (SSC) and was to be completed without a calculator or Data Booklet. Each question had four possible responses with credit awarded for correct answers and no credit deducted for incorrect answers. This paper was assembled from questions submitted by experienced IB teachers or other individuals familiar with the programme.

Teachers' impressions of this paper were conveyed by the 101 G2 forms that were returned. In comparison with last year's paper, 10% thought this year's paper was a little easier, 68% felt it was of a similar standard and 20% considered it a little more difficult. 93% of the respondents thought the level of difficulty was appropriate. Syllabus coverage was considered good by 42%, satisfactory by 56% and poor by 2% of respondents. Clarity of wording was regarded as good by 44%, satisfactory by 54% and poor by 2% of respondents. The presentation of the paper was considered good by 59% and satisfactory by 41% of respondents.

#### Strengths and weaknesses in individual questions

The Difficulty Index (the percentage of candidates achieving each correct answer) ranged from 93% to 24%.

The following comments are made on individual questions.

#### QUESTION 2

Many candidates chose A (which has the greatest **molecular** formula mass), as a result of overlooking the word **empirical** in the question.

#### QUESTION 5

The majority of candidates chose B, which would have been correct if the acid had been monoprotic.

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**Yi-Tong Ma** 

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