



Acid-Base Titrations

Open-Answered: 100%

Name: _____
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A titration is characterized by a point at the equivalence and a neutralization point. Both are necessarily following the stoichiometric ratio of the reaction between an acid and a base. The titration curve is a plot of pH versus volume of titrant. The titration curve of an acid is the opposite of the titration curve of a base.

Titration Calculation Strategies



Example:

When 10.0 mL of a 0.100 M solution of a weak acid (acetic acid) is titrated with 10.0 mL of a 0.100 M solution of a strong base (sodium hydroxide), the pH of the solution is 4.75.

$$\text{CH}_3\text{COOH} + \text{NaOH} \rightarrow \text{CH}_3\text{COO}^- + \text{H}_2\text{O}$$

Calculate the initial concentration of the weak acid (acetic acid) and the initial volume of the strong base (sodium hydroxide) solution.

$$\frac{\text{moles of CH}_3\text{COOH}}{\text{volume of CH}_3\text{COOH}} = \frac{\text{moles of NaOH}}{\text{volume of NaOH}} \quad \text{or} \quad \frac{\text{moles of CH}_3\text{COOH}}{\text{volume of CH}_3\text{COOH}} = \frac{\text{moles of NaOH}}{\text{volume of NaOH}}$$

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Answer the following questions. When you need to use a calculator, use the calculator.

- Calculate the initial concentration of the weak acid (acetic acid) and the initial volume of the strong base (sodium hydroxide) solution.
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web bubbles and breathing once your child has mastered kicking teach them to blow bubbles in the water you can even do this in the bath at home ask them to hold their breath and place their mouth in the water gradually moving to the mouth and nose together many children may get flustered with this lesson

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web even then it s important to be aware of the risks for example a baby may easily swallow too much water which could lead to water poisoning or have trouble adjusting to the cold temperatures in the pool and in rare cases experience hypothermia for children ages 1 to 4 seek parent and child aquatics programs many local community centers

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