# **Acid-Base Titration**

10

A. Concentration of Acetic Acid in Vinega	A.	Concen	tration	of Acetic	Acid i	n Vinegar
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1. Brand Hein Volume 5.00 mL (% on label) 5.0 %

2. Molarity (M) of NaOH \_\_\_\_\_\_ M

	Trial 1	Trial 2	Trial 3
3. Initial NaOH level in buret	G-IML	0.2mL	0.1m1
4. Final NaOH level in buret	396mL	41.2ml	40.101
5. Volume (mL) of NaOH used	Hope to report 1	3115) 3120 150	DOMESTIC CONTRACTOR
6. Average volume (mL)			
7. Average volume in liters (L)	ESTILL OF	72	

# Show all calculations for the following questions.

8. Moles of NaOH used in titration \_\_\_\_\_\_ mole NaOH

9. Moles of HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub> neutralized by NaOH\_\_\_\_\_\_ mole HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>

10. Molarity of HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub> — M HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>

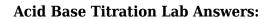
missolled sell chine dominate sell all configurate bluew sell eximine males illegate and existing the sell chine.

11. Grams of HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub> g HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>

12. Percent (m/v) HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub> in vinegar \_\_\_\_\_\_ % HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>

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**Lingjun Ying** 



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