

- 1** Lead(IV) oxide is made when lead reacts with oxygen according to the reaction shown in Equation 1.1.



- a i** Identify, using oxidation numbers, the element that has been reduced.

.....  
..... ( 1 mark)

- ii** Explain, in terms of electrons, what is meant by *reduction*.

.....  
..... ( 1 mark)

- b** Lead(IV) oxide reacts with ice cold hydrochloric acid to produce lead(IV) chloride and water.

- i** Construct an equation for this reaction.

..... ( 1 mark)

- ii** If the reaction is done at higher temperatures, the lead(IV) chloride decomposes to give lead(II) chloride and chlorine. Write an equation for the decomposition of lead(IV) chloride.

..... ( 1 mark)

- 2** Two students are working together to prepare a 250 cm<sup>3</sup> standard solution of sodium carbonate. The method states:

- Using a balance accurate to **two** decimal places, weigh out approximately 1.3 g of anhydrous sodium carbonate accurately.
- Transfer to a small beaker and dissolve the powder in approximately 100 cm<sup>3</sup> of distilled water. Transfer this solution to a 250 cm<sup>3</sup> volumetric flask using a funnel.
- Make up to the mark using distilled water.

The students weighed the sodium carbonate and recorded the mass as 1.4 g.

- a i** Define the term anhydrous.

.....  
..... ( 1 mark)

# Acids Bases And Redox Test Answers Pdf Format

**Horacio J. Adrogué, Donald E. Wesson**



## Acids Bases And Redox Test Answers Pdf Format:

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for the Newport Chemical Agent Disposal Facility As the facility design is being finalized March 1999 April 2000 the NRC will probably be asked to assess all aspects of the facility design including monitoring containment process control and redundancy as well as the quantitative risk assessment QRA This report outlines the elements of the proposed neutralization SCWO technology evaluates the results of ongoing SCWO tests and makes recommendations concerning aspects of the technology that require further development The scope of this evaluation did not include evaluations of other potential technologies or management options for the treatment of VX hydrolysate

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