

PRE-LAB ASSIGNMENT

1. Determine the mass of $\text{H}_2\text{C}_2\text{O}_4 \cdot 2 \text{H}_2\text{O}$ needed to make 250 mL of 0.080 M $\text{H}_2\text{C}_2\text{O}_4$.

2. Determine the volume of 6.0 M NaOH needed to make 500 mL of 0.10 M NaOH.

3. If 32.96 mL of a NaOH solution are required to titrate 25.00 mL of 0.08121 M $\text{H}_2\text{C}_2\text{O}_4$ to the end point, what is the molarity of the sodium hydroxide solution?

4. If 24.92 mL of the NaOH solution from Question 3 are required to titrate 0.1932 g of an unknown diprotic ($n = 2$) acid, what is the molar mass of the acid?

5. In the standardization of the NaOH solution, why is it important to rinse the wet pipet with small portions of the standard oxalic acid solution?

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**United States. Department of
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Services**



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Keywords Index to U.S. Government Technical Reports United States. Department of Commerce. Office of Technical Services,1962 *Some Studies on Acid-base Behaviour in Artificial Seawaters* Andrew Gilmore Dickson,1977

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Report - Food Investigation Board Great Britain. Food Investigation Board,1919 General account of the various activities of the Food Investigation Organization during the year under review divided into 2 parts Report of the Board surveying the main developments during the year and Report of the Director Also includes the reports of its research stations 1931 57 Torrey Research Station on fish Low Temperature Research Station on meat eggs poultry and plant tissues and Ditton Laboratory on fruit and vegetables **Keywords Index to U.S. Government Technical Reports (permuted Title Index).**

United States. Department of Commerce. Office of Technical Services,1962 *Chemical Sensors ... : Proceedings of the Symposium* ,1999 *Index to the Literature of Food Investigation* ,1929 **Energy Research Abstracts** ,1989

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