

Properties of Water Lab

Equipment:

Lab worksheet
Pencil/pen
Penny
Eye dropper
Water

Objective:

Find out how many drops of water can fit on the face of a penny.



Instructions:

- 1- Using the eye-dropper, count the number of drops of water that fit on to the face of the penny. Write down the number of drops it takes. Then change to the other side of the penny and repeat this process.

*Each person will do the experiment three (3) times. (Three Trials)

- 2- Take an average of your three trials and document the data.

	Trial			Average
Heads	1	2	3	XXXXXXXXXX
Number of drops				
Tails	1	2	3	XXXXXXXXXX
Number of drops				

- 3- Write a lab report on the back of this paper about your findings. This report should consist of (4) four paragraphs;

1. Introduction- Hypothesis: What are you trying to discover? What is the purpose of the experiment?

2. Procedures- What was the design of the experiment? What did you do?

3. Data- What did you discover? What does that mean? Use terminology used in class about the properties of water.

4. Conclusion- Results. Include in the conclusion section how this relates to properties of water; and, explain about these properties.

Lab reports will be graded based on content, structure and grammar.

*Remember a paragraph must be at least (4) four sentences long

Properties Of Water Lab Answers

Katrin Zwirglmaier



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Prentice Hall Science Explorer: Teacher's ed ,2005 **Small Water System Operation and Maintenance** ,1987
Integrated Science Laboratory Manual Michael J. Padilla,2000 Includes 74 investigations pre lab discussions and critical thinking questions safety manual and student safety test teaching support **Chemistry in the Community**. American Chemical Society,2002 This volume has relevance to a wide number of courses giving a hands on introduction to chemistry in relation to community issues rather than around specific chemical concepts Activities in Science ,1990 **Selected Water Resources Abstracts** ,1982-10 **Merrill Earth Science** Ralph M. Feather,Susan Leach Snyder,Dale T. Hesser,1995 Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Proceedings of the ... National Symposium on Aquifer Restoration and Ground Water Monitoring** ,1983 **Chemist and Druggist** ,1925 **Laboratory Manual** Arthur H. Brownlow,Mahlon J. Reinhard,1965 *Practical Chemistry* Newton Henry Black,James Bryant Conant,1920
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