

Post Laboratory Assignment

Answer the following on a separate sheet of paper.

- Using the data collected for PF_5 , SF_6 and ClF_3 , what structural comments can be made about molecules with more than four groups (charge clouds) around the central atom?
- Select which structure is preferred and discuss why?



- To assess the stability of the axial vs. equatorial positions for a trigonal bipyramidal structure, a computer program was used to determine the energy for SF_4 with the lone pair in either position (axial or equatorial). The data obtained is shown below:

Equatorial: Energy = -795.14 au

Axial: Energy = -795.088 au

- Using the conversion factor $627.5096 \text{ kcal} = 1 \text{ au}$, what is the energy difference in kcal between the two structures?
 - According to these data, which structure is more stable (more negative)?
 - Does this agree with your explanation from the previous question? Why or why not?
- Below is a table that lists dipole moments and bond angles calculated by a computer program for three molecules. Discuss why the bond angles differ and describe the significance for the dipole moment difference.

Molecule	Dipole moment	Bond Angle
Methane	0.00 Debye	109.47°
Ammonia	1.84 Debye	107.57°
Water	2.15 Debye	105.93°

- Below is a table that lists dipole moments for three molecules. Discuss the ionic and covalent character of each compound using the dipole moment values.

Molecule	Dipole moment
H_2	0.00 Debye
HF	1.94 Debye
LiH	5.95 Debye

Experiment 11 Molecular Models Answers

**J. M. Drake, Joseph Klafter, Raoul
Kopelman**



Experiment 11 Molecular Models Answers:

Energy Research Abstracts ,1986 **Security and Intelligent Information Systems** Pascal Bouvry,Mieczyslaw A. Klopotek,Franck Leprevost,Malgorzata Marciniak,Agnieszka Mykowiecka,Henryk Rybinski,2012-02-02 This book constitutes the thoroughly refereed post conference proceedings of the Joint Meeting of the 2nd Luxembourg Polish Symposium on Security and Trust and the 19th International Conference Intelligent Information Systems held as International Joint Conference on Security and Intelligent Information Systems SIIS 2011 in Warsaw Poland in June 2011 The 29 revised full papers presented together with 2 invited lectures were carefully reviewed and selected from 60 initial submissions during two rounds of selection and improvement The papers are organized in the following three thematic tracks security and trust data mining and machine learning and natural language processing **A-level Chemistry** E. N. Ramsden,2000 Each topic

is treated from the beginning without assuming prior knowledge Each chapter starts with an opening section covering an application These help students to understand the relevance of the topic they are motivational and they make the text more accessible to the majority of students Concept Maps have been added which together with Summaries throughout aid understanding of main ideas and connections between topics Margin points highlight key points making the text more accessible for learning and revision Checkpoints in each chapter test students understanding and support their private study

Molecular Biology Robert Franklin Weaver,2002 This book emphasizes the experimental data and results that support the concepts of molecular biology DNA transcription translation replication and repair Experimental methods are extensively covered The text presumes a prior course in general genetics The text has been updated throughout to reflect the most current ideas tools mechanisms and data used to interpret genetic processes and there is new material on genomics and DNA damage and repair With this new edition all line art which is in full colour tables and many photographs are available as jpeg files on the Visual Resource Library CD ROM and text specific website Scientific and Technical Aerospace Reports ,1993

Chemists' Guide to Effective Teaching Norbert J. Pienta,Melanie M. Cooper,Thomas J. Greenbowe,2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry Journal of Chemical Education ,1924 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers *Dynamics in Small Confining Systems* J. M. Drake,Joseph Klafter,Raoul Kopelman,1990 **What is Heat ?** Frederick Hovenden,1894 **Dekker Encyclopedia of Nanoscience and Nanotechnology** James A. Schwarz,Cristian I. Contescu,Karol Putyera,2004 Scientific American ,1873 Monthly

magazine devoted to topics of general scientific interest

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Engineering and Mining Journal ,1899

Government Reports

Announcements & Index ,1988

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The Electrical World ,1883

Landmark Papers F. C.

Hawthorne,2006 The question of what is matter has fascinated the human race for thousands of years and continues to fascinate us today what is it made of and how does it behave Early in our history the character of natural materials was of critical importance to us and it is no accident that we date the prehistory of humanity by the materials with which our predecessors made their tools Tools are one of the more enduring creations of our prehistoric ancestors and are of particular historical significance as they document the increasing technological sophistication of the human race From the Stone Age to the Bronze Age to the Iron Age there was an increasing awareness of the diversity of natural materials how they could be used and eventually how they could be processed in order to provide even more technologically effective materials for our use This increasing reliance on rocks and minerals required that more and more people be conversant with these materials and their properties The atomistic theory of the Greeks was a solely philosophical construct and further development had to await a more sophisticated approach to Science The first steps in this direction were taken by who else but Isaac Newton 1643 1727 AD Although his ideas on action at a distance initially referred to planets he also considered them as applying to atoms and concluded from physical evidence involving surface tension and viscosity that there must be strong attractions between atoms In what must be considered as insight of legendary proportions Roger Joseph Boscovich 1711 1787 a Jesuit mathematician from Croatia proposed that at very short distances atoms repulse each other the repulsion increasing indefinitely as the particles become closer together whereas at longer distances apart atoms oscillate between attraction and repulsion Frank Hawthorne uses the republication of this set of landmark papers as a vehicle to focus on the development of key issues concerning structural connectivity in inorganic solids of which minerals are a key component and to look at where we are today in our understanding of crystal structure

Metals Abstracts ,1982

ASM News ,2000

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