

Chapter 5: REVIEW: The Periodic Table and Periodicity

Directions: Answer each of the following questions. No need for complete sentences.

- Who first published the classification of the elements that is the basis of our periodic table today?
DIMITRI MENDELEEV
- By what property did Mendeleev arrange the elements?
ATOMIC MASS
- By what property did Moseley suggest that the periodic table be arranged?
ATOMIC NUMBER
- What is the periodic law?
THE PROPERTIES OF THE ELEMENTS REPEAT PERIODICALLY
- What is a period? How many are there in the periodic table?
A HORIZONTAL ROW IN THE PERIODIC TABLE: 7
- What is a group (also called a family)? How many are there in the periodic table?
A VERTICAL COLUMN IN THE PERIODIC TABLE: 18
- State the number of valence electrons in an atom of:

a. sulfur	b. calcium	c. chlorine	d. arsenic
6	2	7	5
- Give the names and chemical symbols for the elements that correspond to these atomic numbers:

a. 19	b. 18	c. 36	d. 90
Ka. POTASSIUM	Ar. ARGON	Kr. KRYPTON	Th. THORIUM
- List, by number, both the period and group of each of these elements:

	Symbol	Symbol	Group
a. beryllium	Be	2	2
b. iron	Fe	8	8
c. lead	Pb	6	14
- Which of the following pairs of elements belong to the same period?
a. He and Cl b. Ne and Li c. He and Cu d. He and Ne
- Which of the following pairs of elements belong to the same group?
a. H and He b. Li and Be **c. C and Pb** d. Ge and Se
- How does an element's period number relate to the number of the energy level of its valence electrons?
PERIOD NUMBER = ENERGY LEVEL OF VALENCE ELECTRONS

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Michael Brown



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