

Acid Base Worksheet

1. Identify whether each is an acid or base:

- | | |
|--------------|-----------------------------|
| <u>acids</u> | Turns blue litmus paper red |
| <u>bases</u> | Turns red litmus paper blue |
| <u>acids</u> | Tastes sour |
| <u>bases</u> | Tastes bitter |

2. A Bronsted-Lowry acid is an proton donor.
3. A Bronsted-Lowry base is an proton acceptor.
4. An Arrhenius acid increases $[H_3O^+]$.
5. An Arrhenius ~~acid~~ base increases $[OH^-]$.

Bronsted-Lowry Worksheet

1. In the following reactions, identify whether H_2O behaves as a Bronsted acid or base.

- a. $HClO_4 + H_2O \rightarrow H_3O^+ + ClO_4^-$ base.
- b. $H_2O + SO_3^{2-} \rightarrow HSO_3^- + OH^-$ acid

2. In the following reactions, circle the Bronsted acid and box the Bronsted base.

- a. $\boxed{CH_3NH_2} + H_2O \rightarrow CH_3NH_3^+ + OH^-$ donor
- b. $\boxed{HSO_4^-} + \boxed{H_2O} \rightarrow H_3O^+ + SO_4^{2-}$ base
- c. $\boxed{HC_2H_3O_2} + \boxed{OH^-} \rightarrow C_2H_3O_2^- + H_2O$
- d. $\boxed{H_2PO_4^-} + \boxed{CO_3^{2-}} \rightarrow HCO_3^- + HPO_4^{2-}$
- e. $\boxed{H_2O} + \boxed{SO_3^{2-}} \rightarrow OH^- + HSO_3^-$

3. Fill in the following table with the appropriate conjugate acid or base:

ACID	CONJUGATE BASE	BASE	CONJUGATE ACID
H_2O	OH^-		
H_2SO_3	HSO_3^-		
		H_2O	H_3O^+
		HPO_4^{2-}	$H_2PO_4^-$

Chapter 8 Solutions Acids Bases Worksheet

**Vibha Arora, Anju Sachdeva, Sushma
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division the game was released for playstation 4 windows and xbox one in october 2019 with a nintendo switch version

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