

## CHEMICAL NAMES AND FORMULAS

### A. Matching

Match each description in Column B to the correct term in Column A. Write the letter of the correct description in the blank.

## Column B

- |                                      |  |
|--------------------------------------|--|
| _____ 1. cation                      | a. any atom or group of atoms with a negative charge   |
| _____ 2. anion                       | b. When two elements combine to form more than one compound, the different masses of one element that combine with the same mass of the other element are in the ratio of small whole numbers. |
| _____ 3. law of definite proportions | c. tightly bound group of atoms that behaves as a unit and carries a charge  |
| _____ 4. acid                        | d. a compound that produces hydroxide ions when dissolved in water   |
| _____ 5. base                        | e. a compound composed of two different elements   |
| _____ 6. law of multiple proportions | f. any atom or group of atoms with a positive charge   |
| _____ 7. polyatomic ion              | g. In all samples of the same chemical compound, the elements are always combined in the same proportion by mass.  |
| _____ 8. monatomic ion               | h. a compound that produces hydrogen ions when dissolved in water  |
| _____ 9. binary compound             | i. consists of a single atom with a positive or negative charge  |

Choose the best customer and write its letter on the line.

- \_\_\_\_\_ 10. The correct name for the  $\text{N}^{3-}$  ion is the:
- a. nitrate ion.  
b. nitric ion.  
c. nitride ion.  
d. nitrite ion.

# Chapter 9 Chemical Names And Formulas Test B Answers

**L Reisser**



## Chapter 9 Chemical Names And Formulas Test B Answers:

Ebook: Physical Science Tillery, 2016-04-16 Ebook Physical Science General, *Organic and Biochemistry* Kate Brown, William Henry Brown, 1983 **Meyer Brothers Druggist**, 1908 **Student's Guide to Introduction to Chemical Principles by Edward I. Peters, 2d Ed** Peter P. Berlow, 1978 Chemistry Charles H. Corwin, 1994 The book focuses on the concepts of chemistry and the applications that maintain and generate motivation for the subject of chemistry

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forensics and more medical related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance 2024-25

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Prentice Hall Chemistry ,2000 **Merrill Chemistry** Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998

**Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science** ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Study Guide** Joan I. Senyk,Darrell D. Ebbing,1984

*Introduction to General, Organic and Biochemistry* M. Scovell,2000-08 To accomplish your course goals use this study guide to enhance your understanding of the text content and to be better prepared for quizzes and tests This convenient manual helps you assimilate and master the information encountered in the text through the use of practice exercises and applications comprehensive review tools and additional helpful resources Introductory Chemistry Charles H. Corwin,2011 To help students learn chemical skills and concepts more effectively Introductory Chemistry Concepts and Critical Thinking Sixth Edition highlights the connection between key concepts and key problem solving skills through critical thinking Math and problem solving are covered early in the text Corwin builds your problem solving ability through innovative learning aids and technology formulated to meet your needs This revision retains all the strengths of the previous editions while adding emphasis on conceptual understanding and critical thinking **General Organic Biochemistry** Bettelh,1988

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