

## CHEMISTRY REVIEW: KEY CHAPTER 9 – CHEMICAL NAMES AND FORMULAS

- In a Stock name such as iron III sulfate, the roman numeral tells us **C**
  - How many atoms of Fe are in one formula unit
  - How many sulfate ions can be attached to the iron atom
  - The charge on each Fe ion
  - The total positive charge of the formula unit
- The result of changing a subscript in a correctly written chemical formula is to **c**
  - Change the number of moles represented in the formula
  - Change the charges on the other ions in the compound
  - Change the formula so that it no longer represents the compound
  - Have no effect on the formula
- Derive a generalization for when an acid name will end in the suffix **-ic** or **-ous**  
**If ending is -ic, the anion ended in -ate; if ending is -ous, the anion ended in -ite**
- Derive a generalization for when an acid name will begin with the prefix **hydro-**.  
**If prefix is hydro-, the anion ended in -ide. This is most likely a monatomic ion.**
- Fill in the blanks in the table below

Compound Name	Formula
Aluminum sulfide	$\text{Al}_2\text{S}_3$
Aluminum sulfite	$\text{Al}_2(\text{SO}_3)_3$
Lead II chloride	$\text{PbCl}_2$
Ammonium phosphite	$(\text{NH}_4)_3\text{PO}_3$
Hydroiodic acid	$\text{HI}$
Dinitrogen pentoxide	$\text{N}_2\text{O}_5$
Copper II carbonate	$\text{CuCO}_3$
Sodium sulfide	$\text{Na}_2\text{S}$
Ammonium phosphate	$(\text{NH}_4)_3\text{PO}_4$
Tin IV sulfite	$\text{Sn}(\text{SO}_3)_2$
Nitrous acid	$\text{HNO}_2$
Magnesium perchlorate	$\text{Mg}(\text{ClO}_4)_2$
Iron II nitrate	$\text{Fe}(\text{NO}_3)_2$
Iron III nitrate	$\text{Fe}(\text{NO}_3)_3$
Cobalt II oxide	$\text{CoO}$
Sulfur trioxide	$\text{SO}_3$
Boron pentafluoride	$\text{BF}_5$
Tetraphosphorus octasulfide	$\text{P}_4\text{S}_8$
Tetraiodide nonoxide	$\text{I}_4\text{O}_9$

- Name the following compound and identify as molecular, ionic or acid
  - $\text{Ca}(\text{HCO}_3)_2$  **calcium bicarbonate - I**
  - $\text{Fe}(\text{ClO}_2)_3$  **Iron III chlorite - I**
  - $\text{Fe}(\text{ClO}_4)_2$  **Iron II chlorate - I**
  - $\text{N}_2\text{O}_5$  **dinitrogen pentoxide - M**

# Chapter 9 Chemical Names And Formulas Answers

**Tiffany Howe,Angela Burton**



## Chapter 9 Chemical Names And Formulas Answers:

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## **Table of Contents Chapter 9 Chemical Names And Formulas Answers**

1. Understanding the eBook Chapter 9 Chemical Names And Formulas Answers
  - The Rise of Digital Reading Chapter 9 Chemical Names And Formulas Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 9 Chemical Names And Formulas Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 9 Chemical Names And Formulas Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 9 Chemical Names And Formulas Answers
  - Personalized Recommendations
  - Chapter 9 Chemical Names And Formulas Answers User Reviews and Ratings
  - Chapter 9 Chemical Names And Formulas Answers and Bestseller Lists
5. Accessing Chapter 9 Chemical Names And Formulas Answers Free and Paid eBooks
  - Chapter 9 Chemical Names And Formulas Answers Public Domain eBooks
  - Chapter 9 Chemical Names And Formulas Answers eBook Subscription Services
  - Chapter 9 Chemical Names And Formulas Answers Budget-Friendly Options
6. Navigating Chapter 9 Chemical Names And Formulas Answers eBook Formats

- ePub, PDF, MOBI, and More
- Chapter 9 Chemical Names And Formulas Answers Compatibility with Devices
- Chapter 9 Chemical Names And Formulas Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter 9 Chemical Names And Formulas Answers
  - Highlighting and Note-Taking Chapter 9 Chemical Names And Formulas Answers
  - Interactive Elements Chapter 9 Chemical Names And Formulas Answers
- 8. Staying Engaged with Chapter 9 Chemical Names And Formulas Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 9 Chemical Names And Formulas Answers
- 9. Balancing eBooks and Physical Books Chapter 9 Chemical Names And Formulas Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 9 Chemical Names And Formulas Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 9 Chemical Names And Formulas Answers
  - Setting Reading Goals Chapter 9 Chemical Names And Formulas Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 9 Chemical Names And Formulas Answers
  - Fact-Checking eBook Content of Chapter 9 Chemical Names And Formulas Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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