CHEM 110 General Chemistry I Laboratory Manual

EXPERIMENT - Verification of Charles' L	aw No	me:		
	Day	y: tos	Instructor:	
LABORAT	TORY REPORT	FORM		
Run #	1	2	3	
1. Temperature of hot water bath	98-5	100-2	101-6	_oc
2. Temperature of cold water bath	0-6	(8.3	43-3	_K
3. Volume of water sucked into flask	36-6	24.2	17-1	_mL
4. Volume of water when flask is full	134.4	140.6	147.1	mL
Calculations:				
 Volume of hot air = Volume of water when flask is full 	-			mL
6. Volume of cold air as measured (no. 5. – no. 3.)	-		-	mL
7. Volume of cold air as predicted by Charles' Law				mL
8. Percent difference				-%
9. Average value for percent difference			%	
o Count until for absolute zero			oc.	

Staple your Graph to this Lab Report Form.

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