

Acid-Base Titrations Pre-Lab Questions (10 Points)

Name _____ Lab Day _____

Lab Instructor _____ Date _____

Show calculations in space provided or on a separate sheet of paper.

1. In an experiment, $\overset{\sqrt{1}}{39.26 \text{ mL}}$ of $\overset{\sqrt{1}}{0.1062 \text{ M}}$ NaOH solution was required to titrate $\overset{\sqrt{2}}{37.54 \text{ mL}}$ of unknown acetic acid solution to a phenolphthalein end point. Calculate the molarity of the acetic acid solution, and the percent (by weight) of acetic acid in the solution (assuming its density to be 1.00 g/mL).

$$\begin{aligned} (M_2)(V_2) &= (M_1)(V_1) \\ M_2(37.54 \text{ mL}) &= (0.1062 \text{ M})(39.26 \text{ mL}) \\ M_2 &= \frac{4.1654 \text{ M mL}}{37.54 \text{ mL}} = 0.1110 \dots \end{aligned}$$

0.1111 M 11.11 %

2. What is the proper disposal of all chemicals in this experiment?

Discard all solutions, HCl solution, rinsing solutions, and titration solutions into your waste beaker. If your waste beaker gets full, discard all solutions in the waste-acid base container.

3. List some the hazards associated with sodium hydroxide.

Sodium hydroxide is toxic ~~and~~, corrosive, and can easily absorbed by the skin.

Chemfax Acid Base Titrations Lab Prelab Answers

Albert A Gayle



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Table of Contents Chemfax Acid Base Titrations Lab Prelab Answers

1. Understanding the eBook Chemfax Acid Base Titrations Lab Prelab Answers
 - The Rise of Digital Reading Chemfax Acid Base Titrations Lab Prelab Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemfax Acid Base Titrations Lab Prelab Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemfax Acid Base Titrations Lab Prelab Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemfax Acid Base Titrations Lab Prelab Answers

- Personalized Recommendations
 - Chemfax Acid Base Titrations Lab Prelab Answers User Reviews and Ratings
 - Chemfax Acid Base Titrations Lab Prelab Answers and Bestseller Lists
5. Accessing Chemfax Acid Base Titrations Lab Prelab Answers Free and Paid eBooks
 - Chemfax Acid Base Titrations Lab Prelab Answers Public Domain eBooks
 - Chemfax Acid Base Titrations Lab Prelab Answers eBook Subscription Services
 - Chemfax Acid Base Titrations Lab Prelab Answers Budget-Friendly Options
 6. Navigating Chemfax Acid Base Titrations Lab Prelab Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemfax Acid Base Titrations Lab Prelab Answers Compatibility with Devices
 - Chemfax Acid Base Titrations Lab Prelab Answers Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemfax Acid Base Titrations Lab Prelab Answers
 - Highlighting and Note-Taking Chemfax Acid Base Titrations Lab Prelab Answers
 - Interactive Elements Chemfax Acid Base Titrations Lab Prelab Answers
 8. Staying Engaged with Chemfax Acid Base Titrations Lab Prelab Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemfax Acid Base Titrations Lab Prelab Answers
 9. Balancing eBooks and Physical Books Chemfax Acid Base Titrations Lab Prelab Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemfax Acid Base Titrations Lab Prelab Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Chemfax Acid Base Titrations Lab Prelab Answers
 - Setting Reading Goals Chemfax Acid Base Titrations Lab Prelab Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Chemfax Acid Base Titrations Lab Prelab Answers

- Fact-Checking eBook Content of Chemfax Acid Base Titrations Lab Prelab Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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web urban rural residence frequency of social contacts was also included as explanatory variable results among cohort members 27 5 were part time caregivers and 6 6 were full time caregivers compared to non caregivers full time caregivers tended to be older to be married more likely to be in the lowest household income group to be

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web may 20 2019 planning for an emergency should be a priority for caregivers and their family members many caregivers are concerned about the health of their loved one in case of an emergency and these tips can help ensure care recipients are well cared for in case of an emergency 1 pack an emergency go bag this bag should include your caregiver