# Identifying the 5 Types of Chemical Reactions

Identify each of the reactions below as one of the 5 types of Reactions (Rxn):

- Synthesis Reaction
- Decomposition Reaction
- Single Replacement Reaction
- · Double Replacement Reaction
- Combustion Reaction

# Reactions (identify pattern below)

Name of Reaction Type

- 1.  $2 \underbrace{Na}_{B} + \underbrace{Cl_{2}}_{A} \rightarrow 2 \underbrace{NaCl}_{A}_{B}$  (A & B combine) Ex: synthesis
- Pb + FeSO<sub>4</sub> → PbSO<sub>4</sub> + Fe
- P<sub>4</sub> + 3 O<sub>2</sub> → 2 P<sub>2</sub>O<sub>3</sub>
- 2 NO₂ → 2 O₂ + N₂
- Na<sub>3</sub>PO<sub>4</sub> + 3 KOH → 3 NaOH + K<sub>3</sub>PO<sub>4</sub>
- C<sub>3</sub>H<sub>6</sub>O + 4 O<sub>2</sub> → 3 CO<sub>2</sub> + 3 H<sub>2</sub>O
- MgCl<sub>2</sub> + Li<sub>2</sub>CO<sub>3</sub> → MgCO<sub>3</sub> + 2 LiCl
- C<sub>6</sub>H<sub>12</sub> + 9 O<sub>2</sub> → 6 CO<sub>2</sub> + 6 H<sub>2</sub>O
- CaCO<sub>3</sub> → CaO + CO<sub>2</sub>
- 2 AgNO<sub>3</sub> + Cu → Cu(NO<sub>3</sub>)<sub>2</sub> + 2 Ag
- 2H₂ + O₂ → 2H₂O

# **Chemical Reactions D Practice Problem 5 Answers**

**CL Gary** 

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