B.12 RELATIVE REACTIVITIES OF METALS

Each Investigating Matter activity has been reframed to provide explicit instruction in the concepts and skills of scientific inquiry. The new subheadings reflect the process of scientific investigation as it is actually conducted by scientists.

Preparing to Investigate

In this investigation, you will observe the reactions of the metals copper, magnesium, and sinc with loar different solutions. Each solution contains a particular ration. The solutions you will use are copper(II) sitrate, Cu(NO₃), (containing Cu²); magnesium solute, Mg(NO₃), (containing Mg²); sinc nitrate, Zu(NO₃), (containing Zu²); and silver nitrate, AgNO₃ (containing Ag²).

Before you begin, read Gathening Evidence to learn what you will need to do and note safety productions. Device a systematic procedure that allows you to observe the reaction [if any) between each metal and each of the four ionic solutions. You will conduct each reaction in a separate well of year well plate using 10 drops of the specified solution and a small strip of

actited to insilicate ropper's specific charges. Copper's sector is CayO because it sortises Cu' serie. CuCo Copper (II) seatle in CuCo It specialists Cu' serie.



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Lei Shi

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