

Name _____ Date _____

Chapter 13 Study Guide

Note to the student...these topics are not in the order that they appear in your packet you will have to do some skipping back and forth through the packet to help you fill these out.

- Heat of vaporization
- Heat of fusion
- What is the boiling point?
- How does atmospheric pressure affect boiling?
- What does the normal boiling point mean?
- What are the condition of STP? (Standard Temperature and Pressure)
- What are the properties of liquids?
- What are the properties of gases?
- What are the properties of solids?
- What is the relationship between average kinetic energy and temperature?
- What happens to average kinetic energies during phase changes?
- What is vapor pressure, how does vapor pressure change with a change in temperature?
- What assumptions have to be considered with the kinetic molecular theory of gases?

Chemistry Chapter 13 States Of Matter Study Guide Answers

Kenneth L. Henold



Chemistry Chapter 13 States Of Matter Study Guide Answers:

Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application

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Foundations of College Chemistry Morris Hein, Susan Arena, Cary Willard, 2023. Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students. It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence. The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need. This International Adaptation offers new and updated content with improved presentation of all course material. It builds on the strengths of previous editions including clear explanations and step by step problem solving. The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry. There is new and expanded coverage of polarizing power and polarizability, Fajans rules, collision number and mean free path, abnormal molecular masses and van t Hoff factor and applications of radioactivity.

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