

Q 11: Define thermoplastics and thermosetting polymers with two examples of each.

Solution:

Thermosetting polymers are called polymers which are cross-linked or strongly branched polymers which are hardened during the moulding process. Through heating, they cannot be softened yet again.
Eg: urea-formaldehyde resins, bakelite.

Thermoplastic polymers are called polymers which are linear slightly branched long-chain polymers which can be softened and hardened repeatedly when heated.

Eg: polystyrene, polythene.

Q 12: Write the monomers used for getting the following polymers.

(i) Polyvinyl chloride

(ii) Teflon

(iii) Bakelite

Solution:

(1) Vinyl chloride ($CH_2 = CHCl$)

(2) Tetrafluoroethylene ($CF_2 = CF_2$)

(3) Phenol (C_6H_5OH) and Formaldehyde (HCHO)

Q 13: Write the name and structure of one of the common initiators used in free radical addition polymerisation.

Solution:

Benzoyl peroxide is one of the common initiator's name and structure used in free radical polymerization.



Q 14: How does the presence of double bonds in rubber molecules influence their structure and reactivity?

Solution:

Natural rubber is a linear cis-polyisoprene consisting of double bonds between the two units of isoprene C_2 and C_3 .

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**Thirumalaisamy P.
Velavancorresponding**



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Chemistry Darrell D. Ebbing, 1999 Technical Note, 1929 Student Solution Supplement for Introduction to Organic Chemistry Andrew Streitwieser, Clayton H. Heathcock, 1976 Bulletin. Library Notes Institute of Paper Chemistry (Appleton, Wis.). Library, 1952 **General Chemistry** Wismer, 1997 Industrial & Engineering Chemistry, 1923
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