Matter and Change, pp. 1-4

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chemical properties

17. compound.

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monity transport

20. grante substance

21. basic

metalloid

23. reactants: carbon and oxygen; product: carbon

reactant: mercury(II) oxide; products: mercury and exygen

26.

27.

296 29%

3400

31.

32.

33.

344

physical change

chemical change

physical change

physical change 330.

chemical change

physical change

- In a solid, particles are packed together in relatively fixed positions. The particles vibrate about a fixed point. In i liquid, particles are close together but can flow a 'ound one another. In a gas, particles are at ere it distances from one another, compared to tile particles of liquids and
- 41. The composition of a pure substance is the same throughout and does not vary from sample to sample. Pure substances include elements and compounds. A mixture contains more than one substance and can vary in composition from one sample to another and s ometimes within different parts of the same sampie.
- 42. Although energy can be absorbed or released in a

change, it is not destroyed or created. It simply takes a different form.

- 43. Homogeneous mixtures are uniform in composition; heterogeneous mixtures are not.
- 44. An element cannot be broken down, but water can be decomposed into hydrogen and oxygen by passing an electrical current through it Celectrolywist.

Measurements and Calculations, pp. 5-8

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0.026 g or 2.6 X 10-2 g 22.

200 2.5 X 10" L

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derived units 24

266 conversion factor.

27. 0.0432 kg or 4.32 X 10"7kg

5400 mL or 5.4 X 103 mL 28.

34000 K. 279.

1.065 X 107 J Tech.

3.54 X 10°cm 311.

- A hypothesis is a testable statement that can be 32. used to make predictions and to carry out further experiments. A theory is a broad generalization that explains a body of facts or phenomena.
- Mass is a measure of the quantity of matter. Weight is a measure of the gravitational pull on matter. Mass does not depend on gravitational ACTION OF THE PARTY.
- 34. Models are visual, verbal, or mathematical representations. They are used to explain how phenomena occur or how data or events are
- 35. The graph of two quantities that are directly proportional is a straight line through the origin.

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Chemistry Matter Change Chapter 16 Answer Key

Rose Arny

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Antarctic Communities International Council of Scientific Unions. Scientific Committee on Antarctic Research,1997-08-28

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Chemistry, Student Study Guide John A. Olmsted, Gregory M. Williams, 2002-01-04 This

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