

CHAPTER 9 REVIEW

Stoichiometry**Teacher Notes and Answers****Chapter 9
SECTION 1****SHORT ANSWER**

1. b
2. d
3. a
4. c
5. a. $2\text{N}_2\text{O}(\text{g}) + 3\text{O}_2(\text{g}) \rightarrow 4\text{NO}_2(\text{g})$
b. 4 mol NO_2 :3 mol O_2
c. 15.0 mol
d. True
e. False
6. a. 28.0 g/mol N_2
2.0 g/mol H_2
17.0 g/mol NH_3
b. 3 mol H_2 :1 mol N_2 ; 2 mol NH_3 :1 mol N_2 ;
2 mol NH_3 :3 mol H_2 ; or their reciprocals
7. a. 1 mol NO :1 mol H_2O
b. 3 mol NO :2 mol NH_3
c. 0.360 mol
8. a. 4 mol O_2 :1 mol C_3H_8 ; 3 mol CO_2 :1 mol C_3H_8 ;
2 mol H_2O :1 mol C_3H_8 ;
3 mol CO_2 :4 mol O_2 ; 2 mol H_2O :4 mol O_2 ;
2 mol H_2O :3 mol CO_2 ; or their reciprocals
b. C_3H_8 is 0.5x; O_2 is 2x; and CO_2 is 1.5x

SECTION 2**PROBLEMS**

1. 4.5 mol
2. 200 g
3. 0.53 g
4. 34.8 g
5. a. 60.2 g
b. 42.1 L
6. a. 81 g
b. 2.9 mol
c. 1.3×10^2 g

SECTION 3**PROBLEMS**

1. 88%
2. a. N_2 ; 2.0 mol
b. 8.0 mol
c. 6.4 mol
3. a. 0.10 mol
b. HCl
c. 1.4 g
4. a. 1.26×10^3 g
b. 960. g
c. 6.9×10^2 L

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G Orfield



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