

# IONIC AND METALLIC BONDING

## Chapter Test A

#### A. Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

	Column A	Column B
E	1. electron dot structure	<ul> <li>a. a mixture of two or more elements, at least one of which is a metal</li> </ul>
F	2. ionic compound	<ul> <li>the number of ions of opposite charge surrounding each ion in a crystal</li> </ul>
H	3. valence electron	<ul> <li>the force of attraction binding oppositely charged ions together</li> </ul>
C	4. ionic bond	d. the attraction of valence electrons for positive metal ions
J	5. chemical formula	<ul> <li>a depiction of valence electrons around the symbol of an element</li> </ul>
		f. compound of cations and anions
G	6. halide ion	g. an anion of a halogen
Α	7. alloy	h. an electron in the highest occupied energy level of an atom
1	8. octet rule	<ol> <li>Atoms in most compounds tend to achieve the electron configuration of a noble gas.</li> </ol>
K	9. formula unit	
		<ol> <li>shows the kinds and numbers of atoms in the smallest representative unit of a substance</li> </ol>
В	10. coordination number	k. lowest whole-number ratio of ions in an ionic compound
D	11. metallic bond	ac sowest whose-manner ratio of some in an some compound

### **B. Multiple Choice**

Choose the best answer and write its letter on the line.

 1.2	. How man	y vadence	electrons	does an	atom of	amy e	element is	n Group 6.	A. harve?
	the state of the s					100			

a. 2 c. 6 b. 4 d. 8

# **Chapter 7 Ionic And Metallic Bonding Test Answers**

**Zhe Xu** 

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## **Table of Contents Chapter 7 Ionic And Metallic Bonding Test Answers**

- 1. Understanding the eBook Chapter 7 Ionic And Metallic Bonding Test Answers
  - The Rise of Digital Reading Chapter 7 Ionic And Metallic Bonding Test Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Chapter 7 Ionic And Metallic Bonding Test Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 7 Ionic And Metallic Bonding Test Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chapter 7 Ionic And Metallic Bonding Test Answers
  - Personalized Recommendations
  - Chapter 7 Ionic And Metallic Bonding Test Answers User Reviews and Ratings
  - Chapter 7 Ionic And Metallic Bonding Test Answers and Bestseller Lists
- 5. Accessing Chapter 7 Ionic And Metallic Bonding Test Answers Free and Paid eBooks
  - Chapter 7 Ionic And Metallic Bonding Test Answers Public Domain eBooks
  - Chapter 7 Ionic And Metallic Bonding Test Answers eBook Subscription Services
  - Chapter 7 Ionic And Metallic Bonding Test Answers Budget-Friendly Options

- 6. Navigating Chapter 7 Ionic And Metallic Bonding Test Answers eBook Formats
  - o ePub, PDF, MOBI, and More
  - Chapter 7 Ionic And Metallic Bonding Test Answers Compatibility with Devices
  - Chapter 7 Ionic And Metallic Bonding Test Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Chapter 7 Ionic And Metallic Bonding Test Answers
  - Highlighting and Note-Taking Chapter 7 Ionic And Metallic Bonding Test Answers
  - Interactive Elements Chapter 7 Ionic And Metallic Bonding Test Answers
- 8. Staying Engaged with Chapter 7 Ionic And Metallic Bonding Test Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 7 Ionic And Metallic Bonding Test Answers
- 9. Balancing eBooks and Physical Books Chapter 7 Ionic And Metallic Bonding Test Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 7 Ionic And Metallic Bonding Test Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 7 Ionic And Metallic Bonding Test Answers
  - Setting Reading Goals Chapter 7 Ionic And Metallic Bonding Test Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 7 Ionic And Metallic Bonding Test Answers
  - Fact-Checking eBook Content of Chapter 7 Ionic And Metallic Bonding Test Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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