

Oxidation-Reduction Titrations I: Determination of Oxalate

B. Standardization of KMnO_4

	Trial 1	Trial 2	Trial 3
Mass of $\text{Na}_2\text{C}_2\text{O}_4$			
Weighing bottle initial mass	<u>2.4331</u>	<u>2.4331</u>	<u>2.4331</u>
Weighing bottle final mass	<u>2.6334</u>	<u>2.6358</u>	<u>2.6340</u>
Mass of $\text{Na}_2\text{C}_2\text{O}_4$	<u>0.2003</u>	<u>0.2027</u>	<u>0.2049</u>
Titration volume of KMnO_4			
Final reading	<u>26.4 mL</u>	<u>25.8 mL</u>	<u>16.5 mL</u>
Initial reading	<u>0.00 mL</u>	<u>0.00 mL</u>	<u>0.00 mL</u>
Volume of KMnO_4	<u>26.4 mL</u>	<u>25.8 mL</u>	<u>16.5 mL</u>

Calculations

mMoles of $\text{Na}_2\text{C}_2\text{O}_4$	_____	_____	_____
Volume of KMnO_4	_____	_____	_____

mMoles of KMnO_4	_____	_____	_____
Molarity of KMnO_4	_____	_____	_____

Average molarity _____

Standard deviation (show calculations) _____

Chemfax Lab Answers Oxidation Reduction Titrations

Paul H. L. Walter



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