

## PRE TEST

## Chapter 6 Test-General/Honors Chemistry

## Part 1: Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

C.2.2      a   1. cationC.2.2      g   2. electronegativityC.2.5      j   3. inner transition metalC.2.6      i   4. alkali earth metalC.2.6      h   5. noble gasC.2.5      b   6. representative elementC.2.2      d   7. metalC.2.6      f   8. halogenC.2.5      e   9. transition metalC.2.2      c   10. ionization energy

a. a positively charged ion

b. an element whose highest occupied *s* or *p* sublevels are partially filled

c. the energy required to remove an electron from an atom

d. good conductor of heat and electric current

e. any element whose highest occupied *s* sublevel and a nearby *d* sublevel contain electrons

f. any nonmetal in group 7A

g. the tendency of an atom to attract electrons when it's in a compound (how much an atom wants an electron)

h. an element in which the highest occupied *s* and *p* sublevels are completely filled

i. an element in group 2A

j. an element whose highest *s* sublevel and a nearby *f* sublevel contain electrons

## Part 2: Multiple Choice

Choose the best answer and write its letter on the line.

C.2.6      b   11. The alkali metals do not include

a. Li

b. Ca

c. Na

d. Rb

C.2.6      c   12. The element iodine, I, is a

a. period 5 alkali metal

b. period 4 halogen

c. period 5 halogen

d. period 5 transition metal

# Chemistry Chapter 6 Test

**Katrin Zwirgmaier**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located on the right side of the slide.

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