

## NOMENCLATURE

homologous series

carboxylic acid

ester

## NOMENCLATURE

prefix

side chain group

methyl

$\text{CH}_3$

ethyl

$\text{CH}_3\text{CH}_2$

propyl

$\text{CH}_3\text{CH}_2\text{CH}_2$

fluoro

F

chloro

Cl

bromo

Br

iodo

I

## STRUCTURAL FORMULAE

↳ shows arrangement of atoms.

eg propane: 1-1-1 =  $\text{CH}_3\text{CH}_2\text{CH}_3$

2,3-dimethylpentane =  $\text{CH}_3\text{CH}_2\text{C}(\text{CH}_3)_2\text{CH}(\text{CH}_3)_2$

but-2-ene =  $\text{CH}_3\text{CH}=\text{CHCH}_3$

## EMPIRICAL FORMULAE

↳ simplest whole ratio of atoms in each element in a compound.

eg

benzene:  $\text{C}_6\text{H}_6$  Empirical formula:  $\text{CH}$

# INTRODUCTION TO ORGANIC CHEMISTRY

nitrite

m

etha

4

2

↳ does not show C or H

↳ but does show other atoms

## AQA A LEVEL CHEMISTRY

ketone

alcohol

amide

aldehyde

## TO CORRECTLY LABEL A STRUCTURE

- 1 Apply prefixes
- 2 Give each group a no.
- 3 Separate no's with commas
- 4 prefixes do not affect the

eg

4 H CH<sub>3</sub> H

## GENERAL

→ alkanes =  $\text{C}_n\text{H}_{2n+2}$

→ alkenes =  $\text{C}_n\text{H}_{2n}$

→ alcohols =  $\text{C}_n\text{H}_{2n+1}\text{OH}$

→ halogenoalkanes =  $\text{C}_n\text{H}_{2n+1}\text{X}$  [X = halogen]



eg

propane = 3 carbons

so  $\text{C}_3\text{H}_{2(3)+2} = \text{C}_3\text{H}_8$

# Chemistry Chm3 W Unit 3 A Introduction To Organic Chemistry

**John M. McIntosh**



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*Organic Chemistry* John M. McIntosh,2022-07-18 This book is intended for beginning students both chemistry majors and other students who require it for their program The material is presented in a concise and student friendly way without the inclusion of topics unnecessary at that level A complete section is designed to lead students through the naming of organic compounds in a self taught manner Reactions are grouped by mechanistic type and stereochemistry is emphasized throughout An introduction to the spectroscopic methods used for structure determination is included Problems are included at each stage and new in this edition are complete answers to the problems as well as an introduction to the molecules of nature      *An Introduction to Organic Chemistry* Harold Goldwhite,1972      **Introduction to Organic Chemistry** William

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Introduction to Organic Chemistry William H. Brown,1999-06-20 The second edition of this introductory book emphasizes the interrelation between organic chemistry and the biological and health sciences      Organic Chemistry. Block 3. Reaction Mechanisms ,1980      Organic Chemistry, Global Edition Leroy G. Wade,Jan W. Simek,2022-08-12 For courses in Organic Chemistry Engage students in complex problem solving with unparalleled pedagogy Wade Simek s Organic Chemistry focuses students on the fundamental reasoning and principles of organic chemistry without compromising the conceptual underpinnings and clear explanations needed to prevent memorization The authors hallmark student oriented approach to problem solving directs students to the individual steps of each reaction shows how the steps contribute to the overall reaction and incorporates unique strategies and hints Organized around functional groups the text incorporates complete discussions of mechanisms The two tiered approach to Mechanism and Key Mechanism Boxes concentrates on 20 Key Mechanisms that comprise nearly all of the mechanisms students will encounter Mastering is the teaching and learning platform that empowers you to reach every student By combining trusted author content with digital tools developed to engage students and emulate the office hour experience Mastering personalizes learning and improves results for each student Mastering Chemistry provides an extension of learning allowing students a platform to practice learn and apply

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