

Chemistry Worksheets Class 8 on Chapter 4 Materials Metals and Non metals - Set 1

Q-1: Which of the following is not a characteristic of metals?

- a) Can be drawn into wires
- b) Generally hard
- c) produces stillness when dropped
- d) Shiny appearance

Q-2: Which of the following statements is not correct?

- a) Iron is a metal and it reacts with water vigorously
- b) Reactions of non metals with base are complex
- c) Non metallic oxides are acidic and metallic oxides are basic
- d) Sodium metal cannot be kept in the open air

Q-3: Which of the following metal/non metal is stored in water?

- a) Na
- b) K
- c) S
- d) P

Q-4: In the table below, identify the metal or nonmetal present in them that forms oxides with oxygen, as well as the chemical formula of the acids and bases from their corresponding nonmetal or metal.

S.No	Chemical formula of the acid	Non metal	Chemical formula of the base	Metal
1.	H_3PO_4		Na_2O	
2.		Carbon		Potassium
3.	H_2SO_4		NH_4OH	
4.	HCl			Magnesium
5.		Nitrogen	Fe_2O_3	

Q-5: Fill in the blanks

- a) The metal which is the best conductor of heat and electricity is _____.
- b) Out of tin, lead, copper and mercury, the metals that lie above hydrogen in the activity series are _____ and _____.

Chemistry D Reading Study Work Answers Chapter 4

Theodore L. Flood, Frank Chapin Bray



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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11
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Treatment Plant Operation Kenneth D. Kerri, 1983 *Glencoe Algebra 1* ,2001 *The Journal of Education* ,1907

School ,1928 *A Different Thermodynamics and its True Heroes* Evgeni B. Starikov, 2019-04-01 Modern

thermodynamics is a unique but still not a logically self consistent field of knowledge It has a proven universal applicability and significance but its actual potential is still latent The development of the foundations of thermodynamics was in effect non stop but absolutely no one has any idea about this This book is the first of its kind that will motivate researchers to build up a logically consistent field of thermodynamics It greatly appreciates the actual depth and potential of thermodynamics which might also be of interest to readers in history and philosophy of scientific research The book presents the life stories of the protagonists in detail and allows readers to cast a look at the whole scene of the field by showcasing a significant number of their colleagues whose works have fittingly complemented their achievements It also tries to trigger a detailed analysis of the reasons why the actual work in this extremely important field has in effect gone astray It comprises five chapters and introduces three scientists in the first two chapters which are specifically devoted to the Scandinavian achievements in macroscopic thermodynamics These introductions are novel and call for a detailed reconsideration of the field The third chapter acquaints the readers with their fourth colleague in Germany who was working on the proper link between the macroscopic thermodynamics kinetics and the atomistic representation of matter The fourth chapter brings in their fifth colleague in the United States who could formally infer the famous formula $S = k \ln W$ ingeniously guessed by Ludwig Boltzmann and thus clarify the physical sense of the entropy notion The last chapter summarizes the above mentioned discourses *The Popular Science News and Boston Journal of Chemistry* ,1892 **Resources in Education** ,1991-08

The American School Board Journal William George Bruce, William Conrad Bruce, 1901 *The Annual American Catalogue* ,1898 **Educational Times** ,1887 *The Publishers Weekly* ,1898 **Publishers' Weekly** ,1898 *Journal of Education* ,1882

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