

7-1 Review and Reinforcement

Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative
positive
crisscross

noble gas
electrons
anion

octet
ions
cation

electron configuration
polyatomic
empirical formula

1. An ionic bond is an attraction between oppositely charged ions.
2. Anions have a negative charge.
3. An atom becomes an ion by losing or gaining electrons.
4. The octet rule states that atoms tend to gain, lose, or share electrons in order to acquire a full set of valence electrons.
5. When sodium and chlorine form an ionic bond, both ions acquire the electron configuration of a(n) noble gas.
6. The atoms that make up polyatomic ions are bonded together by covalent bonds.
7. The crisscross method can be used to write the formula for an ionic compound.
8. The empirical formula of a compound denotes the ratio of ions in the compound.

If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line.

losing

true

true

polyatomic

seven

true (released)

9. Calcium becomes a monatomic cation by gaining two electrons.
10. A cation has a positive charge.
11. A binary ionic compound contains only one kind of cation and one kind of anion.
12. Monatomic ions consist of more than one atom.
13. The Lewis dot diagram for chlorine, a group 7A element, has six electrons.
14. A great deal of energy is produced when an ionic compound is formed.

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Volkan Cicek



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Railway Review , 1912 *Groundwater Chemical Kinetics and Fractal Characteristics of Karst Tunnel* Cangsong Li, Jianfang Ding, Yankai Liao, Song Lu, 2019-11-15 The key to the solution of geological hazards such as Karst water inrush and mud burst in tunnel lies in the accurate prediction or detection of Karst and groundwater By means of on site monitoring theoretical analysis and indoor simulation experiments the authors conduct in depth research on the characteristics of water bearing media and their mechanism of action and explored the relevance of Karst morphology Karst groundwater and fractal characteristics An evaluation model of Karst development degree based on hydrochemical kinetic parameters and fractal index of Karst morphology is established Based on the combination of Karst groundwater dynamics hydrochemistry water rock interaction theory and fractal theory the hydrochemical Kinetics and fractal index evaluation technique for Karst development is proposed It provides a new theory and method for improving the accuracy of Karst and groundwater forecasting The research results are of practical and guiding significance to the construction Karst geological disasters prevention and management of various underground projects in Karst areas Engineers and technicians hydrogeological engineering geologists and college students engaged in tunnel and underground engineering will find it valuable

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