C. Properties of solution, colloids, and suspensions

Solutions	Colloids	Suspensions
Homogeneous	Heterogeneous	Heterogeneous
Particle size: 0.01-1 nm; can be atoms, ions, molecules	Particle size: 1-1000 nm, dispersed; can be aggregates or large molecules	Particle size: over 1000 nm, suspended; can be large particles or aggregates
<u>Do not</u> separate on standing	<u>Do not</u> separate on standing	Particles settle out
Cannot be separated by filtration	Cannot be separated by filtration	<u>Can</u> be separated by filtration
<u>Do not</u> scatter light	Scatter light (Tyndall effect)	May scatter light, but are not transparent

D. <u>Determining if a mixture is a true solution</u>, a colloid, or a suspension:



- a) If particles settle or can be filtered out = suspension
- b) If particles DO NOT settle or filter out shine a beam of light (Tyndall effect) through the mixture
 - If the Tyndall effect is observed = colloid
 - If the Tyndall effect is NOT observed = solution

Chemistry 21a Experiment 6 Solutions And Colloids Introduction

Harry Nicholls Holmes

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