

Chapter 2 Chemistry Assessment Answers

2.1 Assessment Answers

1. Intensive and extensive properties
2. Every sample of a given substance has the same chemical composition.
3. Solid, liquid, gas
4. Physical changes are either reversible or irreversible. Reversible changes can be "undone", or reversed. Irreversible changes cannot be undone.
5. Color; sodium chloride is the only white solid listed.
6. Liquids and gases have indefinite shape. The shape of a solid is definite; the shape of a liquid is indefinite.
7. The freezing of mercury is reversible because solid mercury can be melted.
8. Samples of gold and copper can have the same mass and volume (extensive properties). They cannot have the same set of intensive properties because they have different chemical compositions.

2.2 Assessment Answers

11. As heterogeneous or homogeneous
12. Differences in physical properties.
13. A phase is any part of a sample with uniform composition. There is one phase in a homogeneous mixture and two or more phases in a heterogeneous mixture.
14.
 - a. homogeneous
 - b. heterogeneous
 - c. homogeneous
 - d. heterogeneous
15. Both have a uniform composition throughout. A substance has a definite composition; a solution has a variable composition.
16. Filtrations separates solids from a liquid in a heterogeneous mixture. Distillation can separate a liquid from a substance dissolved in the liquid.
17. Add water to dissolve the salt. Pour the resulting mixture onto a piece of closely woven cloth. The sand will remain on the cloth, and the salt solution will pass through. Use evaporation to remove the water from the salt solution, leaving solid salt behind.

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Elizabeth Kean, Catherine Middlecamp



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