

**CHAPTER 3 CHAPTER ASSESSMENT****MATTER AND CHANGE****Reviewing Vocabulary**

Match the definition in Column A with the term in Column B.

**Column A**

- \_\_\_\_\_ 1. The outermost electrons of an atom
- \_\_\_\_\_ 2. The center of an atom
- \_\_\_\_\_ 3. Hot, highly ionized, electrically conducting gas
- \_\_\_\_\_ 4. The smallest particle of an element that retains that element's characteristics
- \_\_\_\_\_ 5. The change from a gas to a liquid
- \_\_\_\_\_ 6. A homogeneous mixture
- \_\_\_\_\_ 7. The change of one or more substances into other substances
- \_\_\_\_\_ 8. The attractive force between two ions of opposite charge
- \_\_\_\_\_ 9. A tiny particle with a negative electrical charge
- \_\_\_\_\_ 10. The average of the mass numbers of an element's isotopes
- \_\_\_\_\_ 11. The spontaneous process through which unstable nuclei emit radiation

**Column B**

- a. atom
- b. atomic mass
- c. chemical reaction
- d. condensation
- e. electron
- f. ionic bond
- g. solution
- h. nucleus
- i. plasma
- j. valence electrons
- k. radioactive decay

In the space at the left, write the word or phrase in parentheses that makes the statement correct.

- \_\_\_\_\_ 12. The combined number of protons and neutrons is the (atomic number, mass number).
- \_\_\_\_\_ 13. An (energy level, isotope) represents the area in an atom where an electron is most likely to be found.
- \_\_\_\_\_ 14. (An ionic bond, A covalent bond) is the attraction of two atoms for a shared pair of electrons that holds the atoms together.
- \_\_\_\_\_ 15. An atom that gains or loses an electron is a charged particle called (an ion, a molecule).
- \_\_\_\_\_ 16. (An element, A compound) is a substance that cannot be broken down into simpler substances.
- \_\_\_\_\_ 17. A tiny particle that has mass and a positive electrical charge is a (neutron, proton).
- \_\_\_\_\_ 18. (Acids, Bases) are solutions characterized by the formation of hydroxide ions ( $\text{OH}^-$ ).
- \_\_\_\_\_ 19. The process of changing from a liquid into a gas is called (evaporation, sublimation).

# Chemistry Matter And Change Chapter 9 Assessment Answers

**Kaplan**



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