

Solutions

$$\left[\Delta S_{mix} = +ve, \Delta G_{mix} = -ve \right] \text{ (Always)}$$

21. Non-ideal solution :- Do not obey Raoult's law.

A-A or B-B \neq A-B (FOA) + Properties :- $\Delta H_{mix} \neq 0, \Delta V_{mix} \neq 0, P_3 \neq P_A^0 X_A + P_B^0 X_B$

→ +ve deviation from Raoult's law :- A-A + B-B $>$ A-B (FOA)
 Properties :- $P_3 > P_A^0 X_A + P_B^0 X_B$ (Ex. Acetone + Alcohol (ethanol); Benzene + methanol)
 (b) $\Delta H_{mix} = +ve$ (Endothermic rxn, heat absorb) (c) $\Delta V_{mix} = +ve$ Acetone + Benzene
 → -ve deviation from Raoult's law :- A-A + B-B $<$ A-B (FOA)
 Properties :- $P_3 < P_A^0 X_A + P_B^0 X_B$ (Ex. Strong acid + water, Acetone + $CHCl_3$, Phenol + aniline, CHX_3 + (O,N) containing compounds)
 (b) $\Delta H_{mix} = -ve$ (Exothermic rxn, heat release) (c) $\Delta V_{mix} = -ve$

22. Azeotropes :- A mixture of two or more components which boils like pure chemical compound and distills over completely at the same temp without change in chemical composition.
 (i) Minimum boiling azeotropes (ii) Maximum boiling azeotropes
 → B.P. is less than B.P. of either component → B.P. is more than B.P. of either component
 Ex. +ve deviation from Raoult's law Ex. -ve deviation from Raoult's law

Note :- $Y_A = \frac{P_A}{P} = \frac{P_A^0 X_A}{P_A^0 X_A + P_B^0 X_B}$ $P = P_A^0 + P_B^0$

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