- Alkanols can be manufactured from alkenes by the initial reaction of alkenes with
 - A. bromine in tetrachloromethane.
 - B. concentrated tetraoxosulphate(VI) acid.
 - c. aqueous potassium tetraoxomanganate(VII).
 - D. sodium hydroxide solution.
- 8. Which of the following statements about the standard hydrogen electrode is not correct?
 - A. The hydrogen gas is at a pressure of 1 atmosphere
 - A solution containing 1 mol dm⁻³ of H⁺(no) ions is used
 - C. A platinum electrode is used
 - D. The temperature is kept at 20 °C
- If 60 g of M combines with 24 g of oxygen, what would the empirical formula of the oxide be?
 f O = 16.0, M = 1201
 - A. MO
 - B. MO₂
 - C. MO₃
 - D. MO.
- The products of the electrolysis of dilute sodium chloride using carbon electrodes are

	Almostic.		C. CHEROCOCOC
A	chlorine	and	sodium.
B.	oxygen	amd	hydrogen.
C.	chlorine	and	hydrogen.
D.	hydrogen	and	oxygen.

- Determine the quantity of electricity used when a current of 0.20 amperes is passed through an electrolytic cell for 60 minutes.
 - A. 12 C
 - B. 120 C
 - C. 360 C
 - D. 720 C

- Oxochlorate(I) acid is used as a bleaching agent because it is
 - A. a weak acid.
 - B. a reducing agent.
 - C. an oxidizing agent.
 - D. a strong acid.
- The IUPAC name for CH₃CHCH₃CHClCHCH₃CH₃CH₃ is
 - 2, 4 dimethyl 3 chlorohexane.
 - 3, 5 dimethyl 4 chlorohexane.
 - C. 4 chloro 3, 5 dimethylhexane.
 - D. 3 chloro 2, 4 dimethylhexane.
- 14. A colourless gas with a pungent smell is evolved when dilute hydrochloric acid is added to a sample of a salt. The gas evolved could turn
 - A. acidified K₂Cr₂O₂ solution colourless.
 - B. acidified KMnO₄ solution colourless.
 - C. Fe(NO₃)₂ solution green.
 - D. Pb(NO₂)₅ paper black.
- 15. If 5.0 g of marble reacts with 25.0 cm³ hydrochloric acid, which of the following combinations has the fastest reaction rate?
 - A. Marble chips and 2.0 mol dur³ HCl_(an)
 - B. Marble chips and 2.5 mol dnr3 HCl_(eq)
 - C. Powdered marble and 2.5 mol dm⁻³ HCl_(aa)
 - D. Powdered marble and 2.0 mol dm⁻³ HCl_(aq)

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Steven Dulan, Advantage Education (Firm)

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