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Colligative properties are properties that depend only on the number of solute particles in solution and not on the nature of the solute particles.

$$P_1 = X_1 P_1^0$$

$$\Delta T_{\rm b} = K_{\rm b} m$$

$$\Delta T_{\rm f} = K_{\rm f} m$$

Osmotic Pressure
$$(\pi)$$

$$\pi = MRT$$

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D. Das

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