

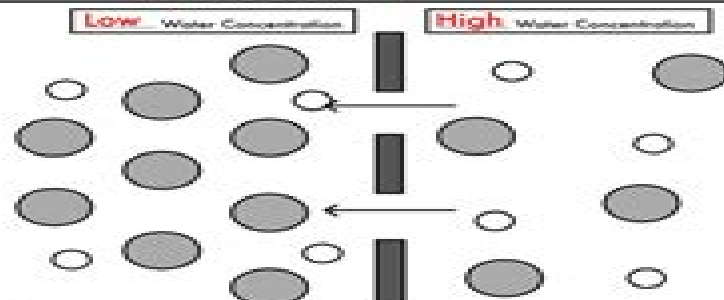
OSMOSIS

Definition of Osmosis

Osmosis is the movement of a substance from an area of **high** water concentration (no/little solute), to an area of **low** water **concentration** (more solute), across a **partially** permeable **membrane**.

Words to use

partially high
membrane low
concentration



KEY: ○ = Water Molecule ● = Solute Molecule

1. Label the two sides of the membrane with either high water concentration or low water concentration.

2. Draw arrows to show the net movement of the water molecules across the membrane.

Effect of Osmosis on Animal Cells

Draw in the boxes below, what the cell would look like in the different conditions.

Cell surrounded by a solution of high water concentration.



Cell swells and bursts (cell lysis)

Cell surrounded by a solution of low water concentration.



Cell shrivels up (becomes crenated)

Effect of Osmosis on Plant Cells

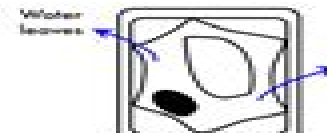
Draw in the boxes below, what the cell would look like in the different conditions.

Cell surrounded by a solution of high water concentration.



Cell swells with water & becomes turgid

Cell surrounded by a solution of low water concentration.



Cell loses water and becomes plasmolysed (cell membrane pulls away from cell wall)

Words to Use

membrane turgid plasmolysed lysis low wall osmosis crenation

Key Points

Animal cells do not have a cell **wall**, if they gain too much water by **osmosis** they will swell and **burst**, known as cell **lysis**. Plant cells on the other hand do have a cell wall and swell to become **turgid**. If animal cells are placed in a solution of **low** water concentration, water will leave the cell causing it to shrivel, known as **crenation**. When plant cells lose water, the cell **membrane** pulls away from the cell wall and is said to be **plasmolysed**.

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