Chemistry I Chapter 5 Study Guide

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- The noble gases, such as belium and xenon, are non-reactive because:
 - a. they have completely filled outer electron shells.
 - they are chemically unstable.
 - c. they are unusually large atoms.
 - d. they have been around longest on the earth.
- Elements in the same column on the periodic table have similar chemical reactivity because:
 - a. they contain similar amounts of energy.
 - they have the same number of valence shell electrons.
 - c. atoms prefer to have completely full or completely empty valence shells.
 - d. they have similar numbers of isotopes.
 - 3. Chlorine, element number 17, is located in the third row and the next-to-the-last group of the periodic table. How many electron shells would be completely filled by a neutral atom of chlorine? How many electrons would be left over?
 - a. I shell filled, 7 electrons left over
 - b. I shell filled. I electron left over
 - c. 2 shells filled, 7 electrons left over
 - d. 2 shells filled. I electron left over
 - 4. How many energy levels are filled in a krypton atom?

36
Kr
Krypton
83.80

- Thirtee.
- From:
- Fireg.
- d. Six.
- 5. The electronegativity of an element is defined as:
 - a. the attraction an atom has for the shared pair of electrons in a covalent bond.
 - the attraction an atom has for the shared pair of electrons in an ionic bond.
 - the amount of static electricity an element produces.
 - d. the amount of charge one atom of an element produces.
- Which of the following is true for valence electrons?
 - Valence electrons are always located in the innermost energy level.
 - Valence electrons may be lost or gained by an atom when bonding occurs with another atom.
 - Each element in the periodic table has a different number of valence electrons.
 - d. Valence electrons are found only in radioactive isotopes.
- Mendeleev arranged the known chemical elements in a table according to increasing
 - a. atomic number.

e. number of protons.

b. number of electrons.

- TTRUCKS.
- 8. Which list of elements contains only metals?
 - a. carbon, iodine, tin

helium, iron, copper

b. tim, copper, cesium.

d. iodine, carbon, argon

- 9. As you move from left to right across a period, the number of valence electrons
 - increases.

increases and then decreases.

b. starys the same.

- d. decreases.
- 10. A member of the boron family has three valence electrons, while a member of the nitrogen family has
 - a. none.

e. four.

Chapter 5 Chemistry Answers

Aftab S. Hassan, James L. Flowers

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