

Science

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(Chapter – 6) (Periodic Classification of Elements) (NIOS)

(Class – X)

Question 2:

Metalloids are present along the diagonal line starting from group 13 and going down to group 16. Do they justify their position in the modern periodic table?

Answer 2:

Location of isotopes: Because the atomic number of all the isotopes of an element is at the same place, they have a place of one in the periodic table.

Intext Questions 6.3

Question 1:

Fill in the blanks with appropriate words

- (a) The force of attraction between nucleus and valence electrons
- (b) Atomic radii of elements in a period from left to right.
- (c) Atomic radii of elements in a group from top to bottom.
- (d) Metallic character of elements from top to bottom in a group.

Answer 1:

- (a) The attraction between the nucleus and the valence electrons **increases** in a period from left to right.
- (b) The atomic radii of elements **decreases** in a period from left to right.
- (c) Atomic radii of elements **increases** in a group from top to bottom.
- (d) Metallic characteristics of elements **increase** from top to bottom in a group.

Question 2:

In the following crossword puzzle, elements are present horizontally, vertically downwards and diagonally downwards. Let us find out how many elements you are able to get within 5 minutes.

Z	N	H	Y	D	R	O	G	E	N
M	B	I	C	A	R	B	O	N	O
A	D	E	T	B	A	R	I	U	M
G	X	Y	H	R	M	U	S	A	S
N	A	D	E	O	O	A	O	O	I
E	I	U	J	P	X	G	I	S	L
S	O	D	I	U	A	M	D	E	I
I	D	M	U	X	A	I	G	N	C
U	I	O	M	O	G	E	Y	R	O
M	N	D	P	S	B	O	R	O	N
A	E	C	H	L	O	R	I	N	E

Please check in the intext answers to find if you missed out any.

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Prentice Hall Exploring Physical Science

Lingjun Ying



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