

Name: _____ Class _____ Period _____ Date _____

Biology Chapter 2 Key Vocabulary

Section 2.1

1. **atom**: the smallest basic unit of matter
2. **element**: one particular type of atom that cannot be broken down by chemical means
3. **compound**: a substance made of atoms of different elements bonded together in a certain ratio
4. **ion**: an atom that gained or lost one or more electrons
5. **ionic bond**: chemical bond formed through the electrical force between oppositely charged ions
6. **covalent bond**: chemical bond formed when atoms share a pair of electrons
7. **molecule**: two or more atoms held together by covalent bonds

Section 2.2

8. **hydrogen bond**: an attraction between a slightly positive hydrogen atom and a slightly negative atom, often oxygen or nitrogen
9. **cohesion**: the attraction among molecules of a substance (ie. water molecules stick to water molecules)
10. **adhesion**: the attraction among molecules of different substances (water molecules sticking to the sides of a glass tube)
11. **solution**: a mixture of substances that is the same throughout (homogeneous mixture)
12. **solvent**: the substance that is present in the greater amount and that dissolves another substance
13. **solute**: a substance that dissolves in a solvent
14. **acid**: a compound that donates a proton when it dissolves in water
15. **base**: a compound that accepts a proton when dissolved in water
16. **pH**: measurement of acidity; related to free hydrogen ion concentration in solution

Section 2.3

17. **monomer**: each subunit in the complete molecule
18. **polymer**: a large molecule, or macromolecule, made of many monomers bonded together
19. **carbohydrate**: molecules composed of carbon, hydrogen, and oxygen; includes sugars and starches
20. **lipid**: nonpolar molecule composed of carbon, hydrogen, and oxygen; includes fats and oils
21. **fatty acids**: chains of carbon atoms bonded to hydrogen atoms
22. **protein**: a polymer made of monomers called amino acids
23. **amino acid**: molecule that makes up proteins, composed of carbon, hydrogen, oxygen, nitrogen, and sometimes sulfur
24. **nucleic acid**: polymer of nucleotides; the genetic material of organisms

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Table of Contents Chapter 3 Science Of Biology Vocabulary Practice Answers

1. Understanding the eBook Chapter 3 Science Of Biology Vocabulary Practice Answers
 - The Rise of Digital Reading Chapter 3 Science Of Biology Vocabulary Practice Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 3 Science Of Biology Vocabulary Practice Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 3 Science Of Biology Vocabulary Practice Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 3 Science Of Biology Vocabulary Practice Answers
 - Personalized Recommendations

- Chapter 3 Science Of Biology Vocabulary Practice Answers User Reviews and Ratings
- Chapter 3 Science Of Biology Vocabulary Practice Answers and Bestseller Lists
- 5. Accessing Chapter 3 Science Of Biology Vocabulary Practice Answers Free and Paid eBooks
 - Chapter 3 Science Of Biology Vocabulary Practice Answers Public Domain eBooks
 - Chapter 3 Science Of Biology Vocabulary Practice Answers eBook Subscription Services
 - Chapter 3 Science Of Biology Vocabulary Practice Answers Budget-Friendly Options
- 6. Navigating Chapter 3 Science Of Biology Vocabulary Practice Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 3 Science Of Biology Vocabulary Practice Answers Compatibility with Devices
 - Chapter 3 Science Of Biology Vocabulary Practice Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 3 Science Of Biology Vocabulary Practice Answers
 - Highlighting and Note-Taking Chapter 3 Science Of Biology Vocabulary Practice Answers
 - Interactive Elements Chapter 3 Science Of Biology Vocabulary Practice Answers
- 8. Staying Engaged with Chapter 3 Science Of Biology Vocabulary Practice Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 3 Science Of Biology Vocabulary Practice Answers
- 9. Balancing eBooks and Physical Books Chapter 3 Science Of Biology Vocabulary Practice Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 3 Science Of Biology Vocabulary Practice Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 3 Science Of Biology Vocabulary Practice Answers
 - Setting Reading Goals Chapter 3 Science Of Biology Vocabulary Practice Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 3 Science Of Biology Vocabulary Practice Answers
 - Fact-Checking eBook Content of Chapter 3 Science Of Biology Vocabulary Practice Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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