

## Chapter 6 Chemical Bonds

**Section 6.2 Covalent Bonding****(pages 166–169)**

This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.

**Reading Strategy** **page 166**

**Relating Text and Visuals** As you read the section, look closely at Figure 5. Complete the table for describing each type of model shown. For more information on this Reading Strategy, see the **Reading and Study Skills** in the Skills and Reference Handbook at the end of your textbook.

| Molecular Models   |             |
|--------------------|-------------|
| Model              | Description |
| Electron dot       |             |
| Structural formula |             |
| Space-filling      |             |
| Electron cloud     |             |

**Covalent Bonds** **(pages 166–169)**

- Describe a covalent bond. \_\_\_\_\_  
\_\_\_\_\_
- Circle the letters of molecular models that show orbitsals of atoms overlapping when a covalent bond forms.
  - electron dot
  - structural formula
  - space-filling
  - electron cloud
- Describe a molecule. \_\_\_\_\_  
\_\_\_\_\_
- Is the following sentence true or false? In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus. \_\_\_\_\_
- Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
  - zero
  - one
  - two
  - three

# Chapter 6 Chemical Bonding Section 2 Covalent Answer Key

**M Mosston**



## Chapter 6 Chemical Bonding Section 2 Covalent Answer Key:

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In a global eaten by displays and the ceaseless chatter of fast conversation, the melodic elegance and emotional symphony created by the prepared term frequently diminish in to the background, eclipsed by the persistent sound and interruptions that permeate our lives. Nevertheless, situated within the pages of **Chapter 6 Chemical Bonding Section 2 Covalent Answer Key** a stunning fictional treasure brimming with fresh thoughts, lies an immersive symphony waiting to be embraced. Constructed by a wonderful musician of language, that captivating masterpiece conducts visitors on a psychological trip, well unraveling the hidden tunes and profound influence resonating within each cautiously crafted phrase. Within the depths of the moving examination, we shall discover the book is main harmonies, analyze its enthralling writing design, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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