

CHEMISTRY 11 – WORD EQUATIONS AND TYPES OF CHEMICAL REACTIONS WORKSHEET

- 1) Translate the following word equations into chemical symbols and balance the resulting equations. Do not include the phases.

(a) Potassium + water \rightarrow potassium hydroxide + hydrogen



(b) Nitrogen + hydrogen \rightarrow ammonia



- 2) In each of the following write a balanced chemical equation, including the phases.

(a) Liquid phosphoric acid reacts with aqueous barium hydroxide to give water and a precipitate of barium phosphate



(b) Solid aluminium oxide and aqueous sulphuric acid produce water and aqueous aluminium sulphate



(c) Nitrogen trifluoride gas and hydrogen gas react to form nitrogen gas and gaseous hydrogen fluoride



(d) Powdered sodium carbonate and aqueous hydrobromic acid react to form carbon dioxide gas, aqueous sodium bromide and water



- 3) Classify the following reactions.

Reaction	Type
$2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$	Combustion
$\text{Mg} + \text{CuSO}_4 \rightarrow \text{MgSO}_4 + \text{Cu}$	Single Replacement
$4\text{Na} + \text{O}_2 \rightarrow 2\text{Na}_2\text{O}$	Synthesis
$\text{Sn}(\text{OH})_4 + 4\text{HBr} \rightarrow 4\text{H}_2\text{O} + \text{SnBr}_4$	Neutralization
$2\text{Fe}(\text{NO}_3)_3 + 3\text{MgS} \rightarrow \text{Fe}_2\text{S}_3 + 3\text{Mg}(\text{NO}_3)_2$	Double Replacement
$2\text{N}_2\text{O} \rightarrow 2\text{N}_2 + \text{O}_2$	Decomposition

Chemical Equations Hand In Assignment 1 Answers

Liying Dong



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Chemical Engineering Progress Symposium Series ,1969 Applied Simulation and Modelling ,1982 A Laboratory Manual of General Chemistry for Use in Colleges William Crowell Bray,Wendell Mitchell Latimer,1921

General Thermodynamics Donald Olander,2007-11-26 Because classical thermodynamics evolved into many branches of science and engineering most undergraduate courses on the subject are taught from the perspective of each area of specialization General Thermodynamics combines elements from mechanical and chemical engineering chemistry including electrochemistry materials science and biology to present a unique and thorough treatment of thermodynamics that is broader in scope than other fundamental texts This book contains classroom tested materials designed to meet the academic requirements for students from a variety of scientific and engineering backgrounds in a single course The first half focuses on classical concepts of thermodynamics whereas the latter half explores field specific applications including a unique chapter on biothermodynamics The book s methodology is unified concise and multidisciplinary allowing students to understand how the principles of thermodynamics apply to all technical fields that touch upon this most fundamental of scientific theories It also offers a rigorous approach to the quantitative aspects of thermodynamics accompanied by clear explanations to help students transition smoothly from the physical concepts to their mathematical representations Each chapter contains numerous worked examples taken from different engineering applications illustrations and an extensive set of exercises to support the material A complete solutions manual is available to professors with qualifying course adoptions

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The Journal of Chemical Physics, 1967. **The Third Conference on Hypercube Concurrent Computers and Applications: Applications** Geoffrey C. Fox, 1988. *Proceedings Parallel Computing*. Fundamental Experiments for College Chemistry Harper Wine Frantz, Lloyd E. Malm, 1963. **Linear Algebra with Applications** Steven J. Leon, 1998. Renowned for its thoroughness and accessibility this best selling text by one of the leading figures in linear algebra reform offers students a challenging yet enjoyable study of linear algebra that is infused with an abundance of applications. Balancing coverage of mathematical theory and applied topics it takes extra care in explaining concepts clearly so that students at a variety of levels can read and understand the material. Numerous worked examples are integrated throughout the text. This revision stresses the important roles played by geometry and visualization in linear algebra. ATLAST Computer Exercises for Linear Algebra a project manual using MATLAB may be packaged free with the text. *Heat Transfer--Philadelphia* W. R. Martini, 1969. Journal of the Ceramic Society of Japan, 1988. Proceedings of the 8th International Conference on Foundations of Computer-Aided Process Design, 2014-07-14. This volume collects together the presentations at the Eighth International Conference on Foundations of Computer Aided Process Design FOCAPD 2014 an event that brings together researchers, educators and practitioners to identify new challenges and opportunities for process and product design. The chemical industry is currently entering a new phase of rapid evolution. The availability of low cost feedstocks from natural gas is causing renewed investment in basic chemicals in the OECD while societal pressures for sustainability and energy security continue to be key drivers in technology development and product selection. This dynamic environment creates opportunities to launch new products and processes and to demonstrate new methodologies for innovation, synthesis and design. FOCAPD 2014 fosters constructive interaction among thought leaders from academia, industry and government and provides a showcase for the latest research in product and process design. Focuses exclusively on the fundamentals and applications of computer aided design for the process industries. Provides a fully archival and

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