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Chemical Bonding

Across

 Bonding which results from the electrical attraction between cations and anions

compound that consists of positive and negative ions

A charged group of covalently bonded atoms

the simplest structual unit of an element

12. ability of a substance to be hammered into thin sheets

 simplest collection of atoms from which an ionic compound's formula can be established

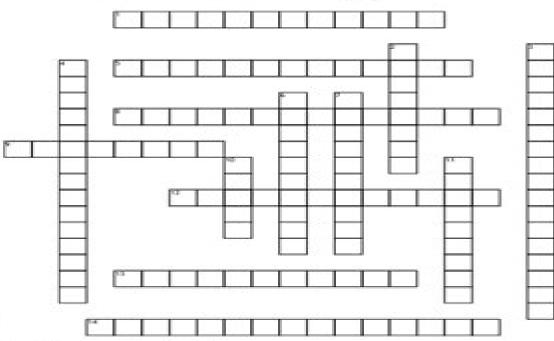
14. when two atoms share a pair of electrons

Down

2. ability of a substance to be drawn or pulled through a small opening to make a wire

Covalent bond when electrons are not shared equally

 bonding that results from the attraction between metal atoms and the surrounding sea of electrons

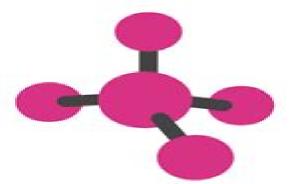


energy required to break a chemical bond and form isolated atoms

covalent bond when one pair of electrons is shared between two atoms

 bonding that has an eneven distribution of charge

 in a chemical compound the max amount of electrons is 8 in the highest energy level



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V. V. Sundaram

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