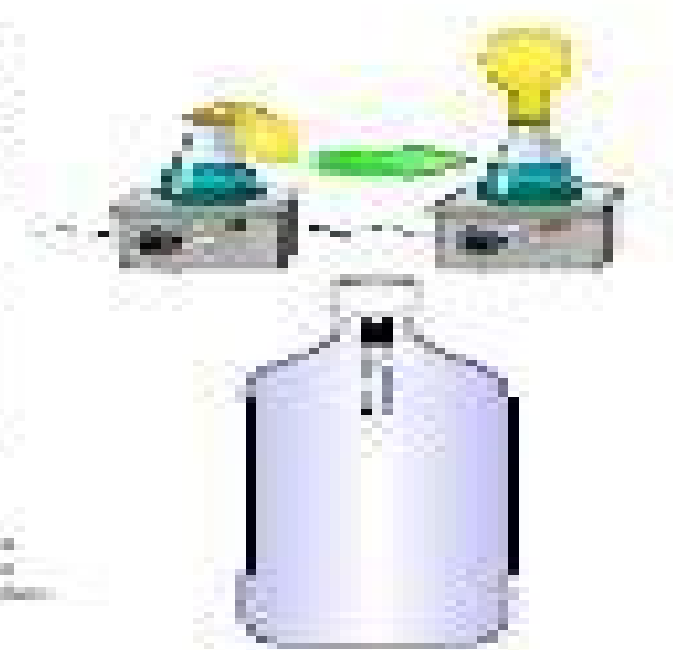


Using the Combined Gas Law

We believe in physics, so we think that the atmosphere is a gas. When the balloon floats up, it is because of the gas. The balloon floats up because the air pressure is lower than the pressure inside the balloon. As the balloon rises, it expands. The air inside the balloon expands, and it pushes against the outside air. This is what causes the balloon to rise. The air pressure is lower than the pressure inside the balloon, so the balloon rises. The air pressure is lower than the pressure inside the balloon, so the balloon rises.



The Law

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Where:

P_1 = initial pressure

V_1 = initial volume

T_1 = initial temperature (K)

P_2 = final pressure

V_2 = final volume

T_2 = final temperature (K)

This equation can be used to find the final pressure, volume, or temperature of a gas if you know the initial pressure, volume, or temperature. It is a useful tool for understanding how gases behave under different conditions.

Example Problem

A gas with a volume of 100 mL at 100 kPa and 300 K is heated to 400 K and the pressure is increased to 150 kPa. What is the final volume?

$$P_1 = 100 \text{ kPa}, T_1 = 300 \text{ K}, P_2 = 150 \text{ kPa}, T_2 = 400 \text{ K}$$

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

$$V_2 = \frac{P_1 V_1 T_2}{P_2 T_1} = \frac{(100 \text{ kPa})(100 \text{ mL})(400 \text{ K})}{(150 \text{ kPa})(300 \text{ K})} = 88.9 \text{ mL}$$

Chemistry Form Ws6 4 1a Answers

Victor M. Corman

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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This book delves into Chemistry Form Ws6 4 1a Answers. Chemistry Form Ws6 4 1a Answers is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chemistry Form Ws6 4 1a Answers, encompassing both the fundamentals and more intricate discussions.

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 - Chapter 3: Chemistry Form Ws6 4 1a Answers in Everyday Life
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