

Answer 1) AQA GCSE C2: Chemistry The Periodic Table Knowledge Answers

This page contains the AQA GCSE Chemistry Periodic Table Questions and their knowledge answers for revision and understanding Periodic Table this page also contains the link to the other end where the revision of this topic.

C 2.1 Development of the periodic table AQA GCSE Chemistry C2 The Periodic Table Knowledge Answers (Page No. 23)

1. Mendeleev's Periodic Table was not perfect as some of the elements did not fit perfectly because the existing elements were not ordered very well. Elements having similar chemical and physical properties were placed in the same row. This showed that some elements were not in the right place in the periodic table.

- a. Gallium and Tellurium are the other two elements which don't quite fit perfectly but do not appear in the order of their relative atomic masses. Gallium has a lower relative atomic mass than Tellurium but comes between them. Tellurium is placed with the chemical properties of other elements.

2. Argon has a greater atomic mass than potassium. According to Mendeleev's periodic table, potassium should come first followed by argon. This resulted in the placement of argon at the end of the first section called the alkali.

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