

IONIC AND METALLIC BONDING

Chapter Test A

A. Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

	Column A	Column B
E	1. electron dot structure	 a. a mixture of two or more elements, at least one of which is a metal
F	2. ionic compound	 the number of ions of opposite charge surrounding each ion in a crystal
H	3. valence electron	 the force of attraction binding oppositely charged ions together
C	4. ionic bond	d. the attraction of valence electrons for positive metal ions
J	5. chemical formula	 a depiction of valence electrons around the symbol of an element
		f. compound of cations and anions
G	6. halide ion	g. an anion of a halogen
Α	7. alloy	h. an electron in the highest occupied energy level of an atom
1	8. octet rule	 Atoms in most compounds tend to achieve the electron configuration of a noble gas.
K	9. formula unit	
		 shows the kinds and numbers of atoms in the smallest representative unit of a substance
В	10. coordination number	k. lowest whole-number ratio of ions in an ionic compound
D	11. metallic bond	ac sowest whose-manner ratio of some in an some compound

B. Multiple Choice

Choose the best answer and write its letter on the line.

 1.2	. How man	y vadence	electrons	does an	atom of	amy e	element is	n Group 6.	A. harve?
	the state of the s					100			

a. 2 c. 6 b. 4 d. 8

Chapter 7 Ionic And Metallic Bonding Test Answers

The Princeton Review

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Table of Contents Chapter 7 Ionic And Metallic Bonding Test Answers

- 1. Understanding the eBook Chapter 7 Ionic And Metallic Bonding Test Answers
 - The Rise of Digital Reading Chapter 7 Ionic And Metallic Bonding Test Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Chapter 7 Ionic And Metallic Bonding Test Answers
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - $\circ \ \ Determining \ Your \ Reading \ Goals$
- 3. Choosing the Right eBook Platform
 - $\circ \ \ Popular \ eBook \ Platforms$
 - Features to Look for in an Chapter 7 Ionic And Metallic Bonding Test Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chapter 7 Ionic And Metallic Bonding Test Answers
 - Personalized Recommendations

- Chapter 7 Ionic And Metallic Bonding Test Answers User Reviews and Ratings
- Chapter 7 Ionic And Metallic Bonding Test Answers and Bestseller Lists
- 5. Accessing Chapter 7 Ionic And Metallic Bonding Test Answers Free and Paid eBooks
 - Chapter 7 Ionic And Metallic Bonding Test Answers Public Domain eBooks
 - Chapter 7 Ionic And Metallic Bonding Test Answers eBook Subscription Services
 - Chapter 7 Ionic And Metallic Bonding Test Answers Budget-Friendly Options
- 6. Navigating Chapter 7 Ionic And Metallic Bonding Test Answers eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Chapter 7 Ionic And Metallic Bonding Test Answers Compatibility with Devices
 - Chapter 7 Ionic And Metallic Bonding Test Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Chapter 7 Ionic And Metallic Bonding Test Answers
 - Highlighting and Note-Taking Chapter 7 Ionic And Metallic Bonding Test Answers
 - Interactive Elements Chapter 7 Ionic And Metallic Bonding Test Answers
- 8. Staying Engaged with Chapter 7 Ionic And Metallic Bonding Test Answers
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 7 Ionic And Metallic Bonding Test Answers
- 9. Balancing eBooks and Physical Books Chapter 7 Ionic And Metallic Bonding Test Answers
 - ∘ Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 7 Ionic And Metallic Bonding Test Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 7 Ionic And Metallic Bonding Test Answers
 - Setting Reading Goals Chapter 7 Ionic And Metallic Bonding Test Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 7 Ionic And Metallic Bonding Test Answers
 - Fact-Checking eBook Content of Chapter 7 Ionic And Metallic Bonding Test Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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