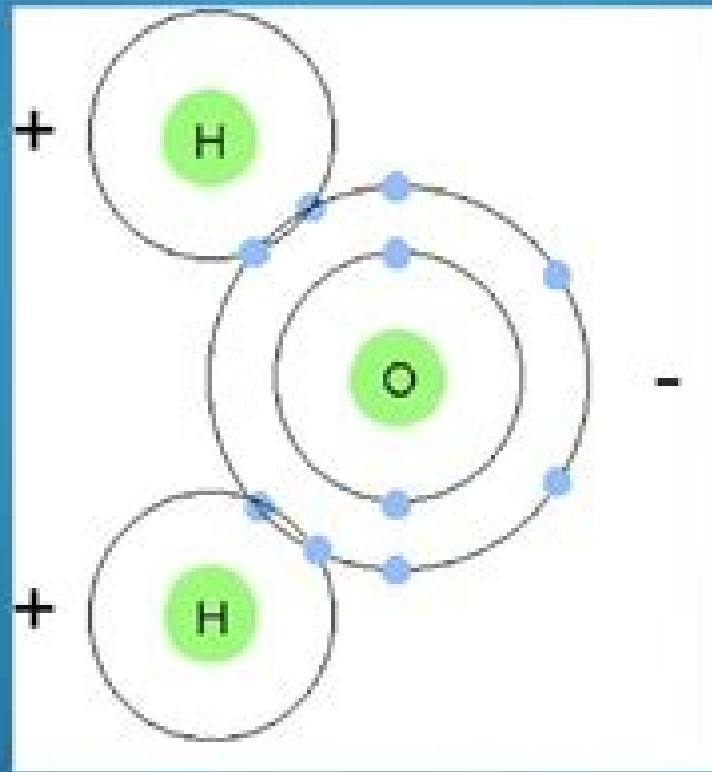


# Chapter 8 – Covalent Bonding



# Chapter 8 Covalent Bonding And Molecular Structure

**John C. Kotz, Paul Treichel, Patrick A.  
Harman**



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Harman,2003 Provides a broad overview of the principles of chemistry the reactivity of chemical elements and their compounds and the applications of chemistry Conveys a sense of chemistry as a field that not only has a lively history but also one that is currently dynamic with important new developments on the horizon Instructor's Manual, General Chemistry, Second Edition Becker,Ralph Sherman Becker,1980 **General Chemistry** James E. Brady,Gerard E. Humiston,1986 The Fifth Edition retains the pedagogical strengths that made the previous editions so popular and has been updated reorganized and streamlined Changes include more accessible introductory chapters with greater stress on the logic of the periodic table earlier introduction of redox reactions greater emphasis on the concept of energy a new section on Lewis structures earlier introduction of the ideal gas law and a new development of thermodynamics Each chapter ends with review questions and problems **Concepts and Models of Inorganic Chemistry** Bodie Eugene Douglas,Darl Hamilton McDaniel,1965 Atomic structure and the periodic table Covalent substances Ionic substances Thermodynamic considerations The hydrogen bond Acids and bases The halogens Electron deficient compounds Metals Coordination compounds structure and bonding stability and reaction mechanisms Experimental methods for the elucidation of structure and bonding of chemical compounds

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