

Section Review 8-1

1. Autotrophs get energy from the sun in the form of light energy. 2. ATP is used to store energy needed for life processes. 3. ADP is converted to ATP by the addition of another phosphate group to an ADP molecule. 4. When ATP is changed to ADP, a phosphate group is removed. The removal of a phosphate group releases energy to the cell. 5. The energy stored in ATP is used for active transport and for movement within cells. 6. They are similar in that they both depend on energy from food. However, autotrophs use the sun's energy to make their own food, while the heterotrophs obtain their energy from the foods they eat. Heterotrophs eat autotrophs or other heterotrophs. 7. heterotroph 8. autotroph 9. heterotroph

Section Review 8-2

1. c 2. b 3. a 4. b 5. c 6. $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$ or carbon dioxide + water \rightarrow sugars + oxygen 7. Light energy is transferred to the electrons in the chlorophyll molecule, raising the energy of these electrons. These high-energy electrons make photosynthesis work. 8. The production of food will drop because plants do not absorb light well in the green region of the visible spectrum. 9. The plant would have leaves of an orange color. The orange pigment carotene absorbs the orange wavelength of light very poorly, which makes the leaves appear orange. 10. Students' experiments should have controls and manipulated and responding variables, and should test an effect of air pollution on photosynthesis. For example, two plants are placed in different atmospheres, one containing few pollutants and the other containing many pollutants. The students would observe the plants and record data on how the plants thrive or survive in different atmospheres.

Section Review 8-3

1. thylakoid 2. Calvin cycle 3. ATP, NADPH 4. oxygen 5. light-independent 6. stroma 7. Both systems are light dependent. Both systems use light energy to energize electrons. One difference is that water molecules are split in photosystem II, but not in photosystem I. In photosystem II, ADP is converted into ATP. 8. The Calvin cycle is a light-independent reaction and does not require light. It will continue as long as there are ATP and NADPH molecules for the cycle to use. 9. The rate of photosynthesis is affected by water, temperature, and light. Low amounts of water, low temperatures, very high temperatures, and little light will slow the rate, while adequate amounts of water, moderate temperatures, and sunny weather will increase the rate. 10. Students should indicate that they would measure either oxygen or carbon dioxide, and that the rate of increase in oxygen or decrease in carbon dioxide would indicate the rate of photosynthesis.

Teaching Resources/Chapter 8

Chapter Vocabulary Review

1. Adenosine triphosphate is one of the principal chemical compounds that living things use to store energy. 2. A thylakoid is a saclike, photosynthetic membrane found in chloroplasts that contains clusters of chlorophyll. 3. NADP⁺ stands for nicotinamide adenine dinucleotide phosphate, a carrier molecule, which accepts and holds 2 high-energy electrons along with a hydrogen ion. 4. ATP synthase is a protein found in the thylakoid membrane and allows H⁺ ions to pass through it. ATP synthase binds ADP and a phosphate group together to produce ATP. 5. The Calvin cycle uses ATP and NADPH from the light-dependent reactions to produce high-energy sugars. The Calvin cycle takes place in the stroma of chloroplasts and does not require light. 6. An autotroph is an organism that makes its own food. A heterotroph is an organism that obtains energy from the food it consumes. 7. Oxygen is produced in the light-dependent reactions of photosynthesis. 8. Pigments are light-absorbing molecules. Chlorophyll is a pigment. 9. The Calvin cycle does not depend on light and uses ATP and NADPH to produce high-energy sugars. The light-dependent reaction uses energy from sunlight to produce ATP, NADPH, and oxygen. 10. Sugars are formed from carbon dioxide in the Calvin cycle. 11. c 12. e 13. b 14. a 15. d 16. light 17. H₂O 18. CO₂ 19. O₂ 20. sugars

Enrichment

1. Radio waves have the longest wavelength on the electromagnetic spectrum. Gamma rays have the shortest wavelength. 2. Boes can see the beginning of the ultraviolet section of the electromagnetic spectrum where the frequencies are close to those of visible light.

Graphic Organizer

1. H₂O 2.–3. NADPH, ATP 4. CO₂ 5.–6. NADP⁺, ADP + P

Design an Experiment

Analyze and Conclude 1. The uncovered leaf and the leaves covered with red and blue cellophane contained starch. The leaves covered with black paper and green cellophane contained little or no starch. Students should explain why their prediction was or was not correct. For example, if students predicted that the leaf covered in black paper would not contain starch, they should explain that the black paper prevented any light from reaching the chlorophyll, so the plant could not make starch in photosynthesis. 2. The blue and red cellophane did not affect photosynthesis. The green cellophane reduced the amount of starch produced by the leaf. Students

Chapter 8 Photosynthesis Section 1 Answer Key

James Morris



Chapter 8 Photosynthesis Section 1 Answer Key:

Primary Processes of Photosynthesis, Part 1 Gernot Renger, 2007-11-29 This volume forms part of a two volume set and is not available for individual purchase Please view the complete pack ISBN 978 0 85404 364 4 for purchase options

Biochemistry of Scandium and Yttrium, Part 1: Physical and Chemical Fundamentals Chaim T. Horovitz, 1999-11-30 Biochemistry of Scandium and Yttrium gathers together existing knowledge about scandium and yttrium from a wide variety of disciplines Part 1 will present a comparative study of the physical and chemical properties of scandium and yttrium looking at both their similarities and their differences Part 2 will address the biochemical aspects of these two elements and the various medical and environmental applications While these elements are relatively rare in nature these books will show that they have unusual physical and chemical properties and a disproportionate number of important applications Improved analytical techniques have revealed that scandium and yttrium are present throughout living matter even though only a relatively limited number of species have been analyzed so far This fact of course has far ranging implications for biological and environmental concerns Part 1 also contains a discussion of the interactions of scandium and yttrium with molecules of biological interest such as organic acids carbohydrates proteins nucleotides and other biologically active molecules The major impacts of scandium and yttrium in science technology and medicine will be of interest to a wide variety of researchers including geochemists inorganic and organic chemists clinical biochemists and those specializing in environmental protection Biochemistry of Scandium and Yttrium Part 1 and Part 2 will be especially welcome because the last book published on the biochemistry of scandium appeared over 20 years ago and the only book mentioning the biochemistry of yttrium came out in 1990

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