

## IONIC BONDING

Ionic bonding occurs when a metal transfers one or more electrons to a nonmetal in an effort to obtain a stable octet of electrons. For example, the transfer of an electron from sodium to chlorine can be depicted by a Lewis dot diagram.



Calcium would need two chlorine atoms to get rid of its two valence electrons.



Show the transfer of electrons in the following combinations.

1.  $\text{K} + \text{F}$



2.  $\text{Mg} + \text{I}$



3.  $\text{Be} + \text{S}$



4.  $\text{Na} + \text{O}$



5.  $\text{Al} + \text{Br}$



## COVALENT BONDING

Name \_\_\_\_\_

Covalent bonding occurs when two or more nonmetals share electrons, attempting to attain a stable octet of electrons at least part of the time. For example:



Note that hydrogen is content with 2, not 8, electrons.

Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

1.  $\text{H} + \text{H} (\text{H}_2)$



2.  $\text{F} + \text{F} (\text{F}_2)$



3.  $\text{O} + \text{O} (\text{O}_2)$



4.  $\text{N} + \text{N} (\text{N}_2)$



5.  $\text{C} + \text{O} (\text{CO})$



6.  $\text{H} + \text{O} (\text{H}_2\text{O})$



# Chapter 8 Covalent Bonding Workbook Answers

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## Unveiling the Power of Verbal Artistry: An Mental Sojourn through **Chapter 8 Covalent Bonding Workbook Answers**

In a world inundated with screens and the cacophony of quick transmission, the profound power and emotional resonance of verbal art often fade into obscurity, eclipsed by the constant assault of sound and distractions. However, nestled within the musical pages of **Chapter 8 Covalent Bonding Workbook Answers**, a charming function of fictional brilliance that impulses with fresh emotions, lies an remarkable journey waiting to be embarked upon. Written by way of a virtuoso wordsmith, this interesting opus books readers on a mental odyssey, delicately exposing the latent possible and profound impact embedded within the delicate internet of language. Within the heart-wrenching expanse of the evocative evaluation, we shall embark upon an introspective exploration of the book is key styles, dissect its captivating publishing model, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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