

CHAPTER 9 REVIEW***Stoichiometry*****MIXED REVIEW****SHORT ANSWER** Answer the following questions in the space provided.

- 1.**
- Given the following equation:
- $C_3H_8(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$

- _____ a. What is the value of the coefficient x in this equation?
- _____ b. What is the molar mass of C_3H_8 ?
- _____ c. What is the mole ratio of O_2 to H_2O in the above equation?
- _____ d. How many moles are in an 8.0 g sample of C_3H_8 ?
- _____ e. If z mol of C_3H_8 react, how many moles of CO_2 are produced, in terms of z ?

- 2.**
- a. What is meant by
- ideal conditions*
- relative to stoichiometric calculations?

- b. What function do ideal stoichiometric calculations serve?

- c. Are actual yields typically larger or smaller than theoretical yields?

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

- 3.**
- Assume the reaction represented by the following equation goes all the way to completion:



- _____ a. If 6 mol of H_2 are consumed, how many moles of NH_3 are produced?
- _____ b. How many grams are in a sample of NH_3 that contains 3.0×10^{23} molecules?

Chapter 9 Mixed Review Stoichiometry Answers

Lingsheng Yao



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **Holt Chemistry** Ralph Thomas Myers, 2004 *Study Guide, Introductory Chemistry, a Foundation ; Introductory Chemistry ; Basic Chemistry* Iris Stovall, Donald J. DeCoste, 2000 **Basic Chemistry** Steven S Zumdahl, 1996 **Thermodynamic Data** Surendra K. Saxena, 2012-12-06 With the rapid development of fast processors the power of a mini super computer now exists in a lap top box Quite sophisticated techniques are becoming accessible to geoscientists thus making disciplinary boundaries fade Chemists and physicists are no longer shying away from computational mineralogical and material science problems too complicated to handle Geoscientists are willing to delve into quantitative physico chemical methods and open those black boxes they had shunned for several decades but with which had learned to live I am proud to present yet another volume in this series which is designed to break the disciplinary boundaries and bring the geoscientists closer to their chemist and physicist colleagues in achieving a common goal This volume is the result of an international collaboration among many physical geochemists chemists physicists and geologists aiming to understand the nature of material The book has one common theme namely how to determine quantitatively through theory the physico chemical parameters of the state of a solid or fluid **Chemical Engineering and Mining Review**, 1915 **The Practice of Chemistry** Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt 832KB **Applied Mechanics Reviews**, 1986 **Chemical Principles** Peter Atkins, Loretta Jones, 2007-08 Written for calculus inclusive general chemistry courses Chemical Principles helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry's frontier continually demonstrating how to solve problems think about nature and matter and visualize chemical concepts as working chemists do Flexibility in level is crucial and is largely established through clearly labeling separating in boxes the calculus coverage in the text Instructors have the option of whether to incorporate calculus in the coverage of topics The multimedia integration of Chemical Principles is more deeply established than any other text for this course Through the unique eBook the comprehensive Chemistry Portal Living Graph icons that connect the text to the Web and a complete set of animations students can take full advantage of the wealth of resources available to them to help them learn and gain a deeper understanding **Introductory Chemistry** Steven S. Zumdahl, 1996 **Chemical Analysis**

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