

**Section 1.1** *continued*

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

19
39
55
87
101
119

Atomic Mass	1.1
Atomic Number	1.2
Electron Configuration	1.3
Chemical Name	1.4
Chemical Symbol	1.5

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 16. It appears in the periodic table.
- _____ 17. It has no line periodic table.
- _____ 18. Group 1 elements.
- _____ 19. Elements that are shiny and conduct electricity.
- _____ 20. Group 8 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ 21. There are two representative elements of elements.
- _____ 22. Most elements have lines in the periodic table for metalloids.
- _____ 23. Group 1A elements except for hydrogen are metals in the solid state.
- _____ 24. Group 2A elements are the alkaline earth metals.
- _____ 25. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 26. Group 8A elements are inert nonmetallic elements known as noble gases.
- _____ 27. Elements have properties of both metals and nonmetals.

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Table of Contents Chemistry Chapter 6 Study Guide For Content Mastery Answers

1. Understanding the eBook Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - The Rise of Digital Reading Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - User-Friendly Interface

4. Exploring eBook Recommendations from Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Personalized Recommendations
 - Chemistry Chapter 6 Study Guide For Content Mastery Answers User Reviews and Ratings
 - Chemistry Chapter 6 Study Guide For Content Mastery Answers and Bestseller Lists
5. Accessing Chemistry Chapter 6 Study Guide For Content Mastery Answers Free and Paid eBooks
 - Chemistry Chapter 6 Study Guide For Content Mastery Answers Public Domain eBooks
 - Chemistry Chapter 6 Study Guide For Content Mastery Answers eBook Subscription Services
 - Chemistry Chapter 6 Study Guide For Content Mastery Answers Budget-Friendly Options
6. Navigating Chemistry Chapter 6 Study Guide For Content Mastery Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Chapter 6 Study Guide For Content Mastery Answers Compatibility with Devices
 - Chemistry Chapter 6 Study Guide For Content Mastery Answers Enhanced eBook Features
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 - Highlighting and Note-Taking Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Interactive Elements Chemistry Chapter 6 Study Guide For Content Mastery Answers
8. Staying Engaged with Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Chapter 6 Study Guide For Content Mastery Answers
9. Balancing eBooks and Physical Books Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Chapter 6 Study Guide For Content Mastery Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Setting Reading Goals Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Fact-Checking eBook Content of Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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