

PRE TEST

Chapter 6 Test-General/Honors Chemistry

Part 1: Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

- | | | | |
|-------|--------------|---------------------------|--|
| C.2.2 | <u> a </u> | 1. cation | a. a positively charged ion |
| C.2.2 | <u> g </u> | 2. electronegativity | b. an element whose highest occupied <i>s</i> or <i>p</i> sublevels are partially filled |
| C.2.5 | <u> j </u> | 3. inner transition metal | c. the energy required to remove an electron from an atom |
| C.2.6 | <u> i </u> | 4. alkali earth metal | d. good conductor of heat and electric current |
| C.2.6 | <u> h </u> | 5. noble gas | e. any element whose highest occupied <i>s</i> sublevel and a nearby <i>d</i> sublevel contain electrons |
| C.2.5 | <u> b </u> | 6. representative element | f. any nonmetal in group 7A |
| C.2.2 | <u> d </u> | 7. metal | g. the tendency of an atom to attract electrons when it's in a compound (how much an atom wants an electron) |
| C.2.6 | <u> f </u> | 8. halogen | h. an element in which the highest occupied <i>s</i> and <i>p</i> sublevels are completely filled |
| C.2.5 | <u> e </u> | 9. transition metal | i. an element in group 2A |
| C.2.2 | <u> c </u> | 10. ionization energy | j. an element whose highest <i>s</i> sublevel and a nearby <i>f</i> sublevel contain electrons |

Part 2: Multiple Choice

Choose the best answer and write its letter on the line.

- | | | | | |
|-------|--------------|--------------------------------------|--------------------------|------------------------------|
| C.2.6 | <u> b </u> | 11. The alkali metals do not include | a. Li | c. Na |
| | | | b. Ca | d. Rb |
| C.2.6 | <u> c </u> | 12. The element iodine, I, is a | a. period 5 alkali metal | c. period 5 halogen |
| | | | b. period 4 halogen | d. period 5 transition metal |

Chemistry Chapter 6 Test

Augustus Baldwin Wadsworth



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