CHEMISTRY 1 FINAL EXAM REVIEW

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Accountry

Name Section Date

Chemistry Review: Writing and Balancing Equations

Directions: Solve each problem and answer each question

- 1. Balance each equation below by adding the correct coefficients
 - a. Goth, + Note + Not + Gotte,
 - b. __HON+_M60H9,-I-_M60N9,+_H,D
 - c. __H90,+__(981(3)50,-9__MH,NO,+__H50,
 - d. _2MDDL+_MO, +_2MO +_MDDL
 - a. __C,H_a + __O₂ -> __OO₂ + __N₂O
- 2. Write and balance each equation, include phase labels.
 - Aguerous sodium hydroxide reacts with aquerous suffuric acid to form aquerous sodium suffare and reader
 - Solid inon (XI) hydroxide and aqueous solitum carbonate are formed from the seartion of solid inon (XI) carbonate and aqueous solitum hydroxide
 - Solid aluminum reacts with aqueous hydrobromic acid to form aqueous aluminum bromide and hydrogen gas
 - Gaseous ethane (dicarbon hesshydride) reacts with oxygen gas to form passous. carbon-dicaids and passous water
 - Solid calcium phosphite reacts with aqueous copper(10 nitrate to firm aqueous calcium nitrate and solid copper (10 phosphite)

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21 CHEMISTRY WORKSHEETS WORKED ANSWER KEY

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