

CHAPTER 12 REVIEW*Liquids and Solids***SECTION 12-2**

SHORT ANSWER Answer the following questions in the space provided.

1. Match the following descriptions on the right to the crystal type on the left.

_____ ionic crystal

(a) has mobile electrons in the crystal

_____ covalent molecular crystal

(b) is hard, brittle, and nonconducting

_____ metallic crystal

(c) typically has the lowest melting point of the four crystal types

_____ covalent network crystal

(d) has strong covalent bonds between neighboring atoms

2. For each of the four types of solids, give a specific example other than those listed in Table 12-1 on page 370 of the text.

3. A chunk of solid lead is dropped into a pool of molten lead. The chunk sinks to the bottom of the pool. What does this tell you about the density of the solid lead compared with the density of the molten lead?

4. Answer *solid* or *liquid* to the following questions:

_____ a. Which is more incompressible?

_____ b. Which is quicker to diffuse into neighboring media?

_____ c. Which has a definite volume and shape?

_____ d. Which has molecules that are primarily rotating or vibrating in place?

Chemistry Chapter 12 Review Liquids And Solids Answers

Arshad Iqbal



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